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Title of Invention: Biosensor

Inventors (please provide full names): \_\_\_\_\_

Earliest Priority Filing Date: \_\_\_\_\_

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L77 ANSWER 1 OF 33 HCAPLUS COPYRIGHT 2003 ACS  
AN 2002:495714 HCAPLUS  
DN 137:246666  
TI Quinohemoprotein alcohol dehydrogenase-based reagentless amperometric biosensor for ethanol monitoring during wine fermentation  
AU Niculescu, Mihaela; Erichsen, Thomas; Sukharev, Valentin; Kerenyi, Zoltan; Csoregi, Elisabeth; Schuhmann, Wolfgang  
CS Department of Biotechnology, Lund University, Lund, S-22100, Swed.  
SO Analytica Chimica Acta (2002), 463(1), 39-51  
CODEN: ACACAM; ISSN: 0003-2670  
PB Elsevier Science B.V.  
DT Journal  
LA English  
CC 17-1 (Food and Feed Chemistry)  
AB This paper describes the development and optimization of an amperometric biosensor for monitoring ethanol in beverages. The biosensor is constructed by crosslinking a quinoprotein alc. dehydrogenase (QH-ADH) to an Os-complex-modified poly(vinylimidazole) redox polymer using poly(ethylene glycol) diglycidyl ether. The optimum biosensor configuration was evaluated by changing the ratio between enzyme, redox polymer, and cross-linker using conventional graphite rods as basis electrodes. The optimized sensor showed a sensitivity of 0.336.+-.0.025 A M-1 cm<sup>2</sup> for ethanol and a detection limit (calcd. as three times the signal-to-noise ratio) of 1 .mu.M. This biosensor configuration was further evaluated in a conventional flow-injection system and the applicability for the detn. of ethanol in diverse wine samples could be successfully demonstrated. Adaptation of this sensor configuration to screen-printed (SP) electrodes allowed their integration into an automated sequential-injection analyzer and the successful online monitoring of ethanol during wine fermn. processes.  
ST ethanol detn wine quinohemoprotein alc dehydrogenase amperometric biosensor  
IT Enzyme electrodes (amperometric; quinohemoprotein alc. dehydrogenase-based reagentless amperometric biosensor for ethanol monitoring during wine

fermn.)

IT Wine  
 (quinohemoprotein alc. dehydrogenase-based reagentless amperometric biosensor for ethanol monitoring during wine fermn.)

IT 26403-72-5, Poly(ethylene glycol)  
 diglycidyl ether  
 RL: MOA (Modifier or additive use); USES (Uses)  
 (crosslinking agent; quinohemoprotein alc. dehydrogenase-based reagentless amperometric biosensor for ethanol monitoring during wine fermn.)

IT 64-17-5, Ethanol, analysis  
 RL: ANT (Analyte); ANST (Analytical study)  
 (quinohemoprotein alc. dehydrogenase-based reagentless amperometric biosensor for ethanol monitoring during wine fermn.)

IT 37205-43-9, Quinohemoprotein alcohol dehydrogenase  
 RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)  
 (quinohemoprotein alc. dehydrogenase-based reagentless amperometric biosensor for ethanol monitoring during wine fermn.)

IT 7440-04-2D, Osmium, poly(vinylimidazole)  
 complexes 25232-42-2D, Poly(vinylimidazole),  
 osmium complexes  
 RL: TEM (Technical or engineered material use); USES (Uses)  
 (quinohemoprotein alc. dehydrogenase-based reagentless amperometric biosensor for ethanol monitoring during wine fermn.)

RE.CNT 25 THERE ARE 25 CITED REFERENCES AVAILABLE FOR THIS RECORD

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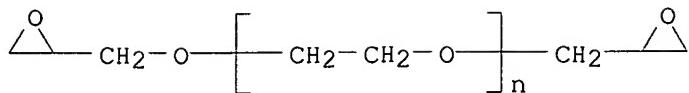
IT 26403-72-5, Poly(ethylene glycol)

diglycidyl ether

RL: MOA (Modifier or additive use); USES (Uses)  
 (crosslinking agent; quinohemoprotein alc. dehydrogenase-based reagentless amperometric biosensor for ethanol monitoring during wine fermn.)

RN 26403-72-5 HCPLUS

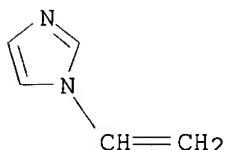
CN Poly(oxy-1,2-ethanediyl), .alpha.- (oxiranylmethyl)-.omega.- (oxiranylmethoxy)- (9CI) (CA INDEX NAME)



IT 7440-04-2D, Osmium, poly(vinylimidazole)  
complexes 25232-42-2D, Poly(vinylimidazole),  
osmium complexes  
RL: TEM (Technical or engineered material use); USES (Uses)  
(quinohemoprotein alc. dehydrogenase-based reagentless amperometric  
biosensor for ethanol monitoring during wine fermn.)  
RN 7440-04-2 HCAPLUS  
CN Osmium (8CI, 9CI) (CA INDEX NAME)

Os

RN 25232-42-2 HCAPLUS  
CN 1H-Imidazole, 1-ethenyl-, homopolymer (9CI) (CA INDEX NAME)  
CM 1  
CRN 1072-63-5  
CMF C5 H6 N2



L77 ANSWER 2 OF 33 HCAPLUS COPYRIGHT 2003 ACS  
AN 2002:283240 HCAPLUS  
DN 137:19560  
TI Redox hydrogel-based bienzyme microelectrodes for  
amperometric monitoring of L-glutamate  
AU Mikeladze, Ekaterina; Schulte, Albert; Mosbach, Marcus; Blochl, Andrea;  
Csoregi, Elisabeth; Solomonia, Revaz; Schuhmann, Wolfgang  
CS Biochemical Neuropharmacology, Institute of Physiology, Georgian Academy  
of Sciences, Tbilisi, 380060, Georgia  
SO Electroanalysis (2002), 14(6), 393-399  
CODEN: ELANEU; ISSN: 1040-0397  
PB Wiley-VCH Verlag GmbH  
DT Journal  
LA English  
CC 17-1 (Food and Feed Chemistry)  
AB Fabrication and characterization of amperometric bienzyme L-glutamate  
sensitive microelectrodes are the prerequisite for monitoring changes of  
L-glutamate concn. at glutamate-secreting cell cultures. The design of  
the glutamate microelectrodes is based on incorporating L-glutamate  
oxidase and horseradish peroxidase into a  
redox-hydrogel contg. PVI19-dmeOs as the redox  
mediator and immobilizing this system onto the surface of platinum  
microdisk electrodes using a dip-coating procedure. For amperometric  
measurements of L-glutamate, these redox hydrogel  
-based bienzyme microelectrodes can be operated at low working potentials  
(-50 mV vs. Ag/AgCl) decreasing the influence of electroactive

interferants possibly present in biol. samples. The L-glutamate microsensors are characterized by a good operation stability and sensitivity (0.038.+-0.005 mAM-1), a low detection limit (0.5 .mu.M in a conventional amperometric set-up and 0.03 .mu.M in a Faraday cage, defined as three times the signal-to-noise ratio), a linear range up to 50 .mu.M and a response time of about 35 s. The glutamate biosensors have been applied for the direct measurement of L-glutamate release (upon chem. stimulation) from a population of immortalized hippocampal neurons (HN10 cells) demonstrating the possibility to amperometrically monitor in-situ L-glutamate secretion from these cells.

ST glutamate amperometric enzyme microelectrode

IT Enzyme electrodes

(amperometric; **redox hydrogel-based bienzyme**  
microelectrodes for amperometric monitoring of L-glutamate)

IT Microelectrodes

(enzyme; **redox hydrogel-based bienzyme**  
microelectrodes for amperometric monitoring of L-glutamate)

IT Brain

(hippocampus; **redox hydrogel-based bienzyme**  
microelectrodes for amperometric monitoring of L-glutamate)

IT Enzyme electrodes

(microelectrodes; **redox hydrogel-based bienzyme**  
microelectrodes for amperometric monitoring of L-glutamate)

IT 9003-99-0, Peroxidase

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)  
(horseradish; **redox hydrogel-based**  
bienzyme microelectrodes for amperometric monitoring of L-glutamate)

IT 56-86-0, L-Glutamic acid, analysis

RL: ANT (Analyte); ANST (Analytical study)  
(**redox hydrogel-based** bienzyme microelectrodes for  
amperometric monitoring of L-glutamate)

IT 39346-34-4, L-Glutamate oxidase

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)  
(**redox hydrogel-based** bienzyme microelectrodes for  
amperometric monitoring of L-glutamate)

RE.CNT 40 THERE ARE 40 CITED REFERENCES AVAILABLE FOR THIS RECORD

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IT 9003-99-0, Peroxidase

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)  
 (horseradish; redox hydrogel-based  
 bienzyme microelectrodes for amperometric monitoring of L-glutamate)

RN 9003-99-0 HCPLUS

CN Peroxidase (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

- L77 ANSWER 3 OF 33 HCPLUS COPYRIGHT 2003 ACS  
 AN 2002:205115 HCPLUS  
 DN 137:108425  
 TI Amperometric enzyme-based biosensors for application in food and beverage industry  
 AU Csoregi, Elisabeth; Gaspar, Szilveszter; Niculescu, Mihaela; Mattiasson, Bo; Schuhmann, Wolfgang  
 CS Centre for Chemistry and Chemical Engineering, Department of Biotechnology, Lund University, Lund, 221 00, Swed.  
 SO Focus on Biotechnology ((2001), 7(Physics and Chemistry: Basis of Biotechnology), 105-129  
 CODEN: FBOIAM  
 PB Kluwer Academic Publishers  
 DT Journal; General Review  
 LA English  
 CC 17-0 (Food and Feed Chemistry)  
 AB A review. Continuous, sensitive, selective, and reliable monitoring of a large variety of different compds. in various food and beverage samples is of increasing importance to assure a high-quality and tracing of any possible source of contamination of food and beverages. Most of the presently used classical anal. methods are often requiring expensive instrumentation, long anal. times and well-trained staff. Amperometric enzyme-based biosensors on the other hand have emerged in the last decade from basic science to useful tools with very promising application possibilities in food and beverage industry. Amperometric biosensors are in general highly selective, sensitive, relatively cheap, and easy to integrate into continuous anal. systems. A successful application of such sensors for industrial purposes, however, requires a sensor design, which satisfies the specific needs of monitoring the targeted analyte in the particular application. Since each individual application needs different operational conditions and sensor characteristics, it is obvious that biosensors have to be tailored for the particular case. The characteristics of the biosensors are depending on the used biorecognition element (enzyme), nature of signal transducer (electrode material) and the communication between these two elements (electron-transfer pathway). Therefore, the present chapter presents the different existing biosensor designs describing the possible electron-transfer pathways, discusses their advantages and disadvantages, and shows their possible application in food and beverage industry. Three practical

examples are given describing **biosensor** designs developed in our lab., demonstrating their usefulness for industrial applications.  
ST review amperometric enzyme **biosensor** food beverage analysis  
IT Beverages  
Food analysis  
Food industry  
Quality control  
(amperometric enzyme-based **biosensors** for application in food and beverage industry)  
IT Biosensors  
(amperometric; amperometric enzyme-based biosensors for application in food and beverage industry)  
IT Biosensors  
(enzymic, electrochem.; amperometric enzyme-based biosensors for application in food and beverage industry)

RE.CNT 91 THERE ARE 91 CITED REFERENCES AVAILABLE FOR THIS RECORD  
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L77 ANSWER 4 OF 33 HCAPLUS COPYRIGHT 2003 ACS  
AN 2002:115443 HCAPLUS  
DN 136:213052  
TI Direct bioelectrocatalysis at carbon electrodes modified with  
quinohemoprotein alcohol dehydrogenase from Gluconobacter sp. 33  
AU Razumiene, J.; Niculescu, M.; Ramanavicius, A.; Laurinavicius,  
V.; Csoregi, E.  
CS Institute of Biochemistry Vilnius, Vilnius, LT-2600, Lithuania  
SO Electroanalysis (2002), 14(1), 43-49  
CODEN: ELANEU; ISSN: 1040-0397  
PB Wiley-VCH Verlag GmbH  
DT Journal  
LA English  
CC 9-7 (Biochemical Methods)  
AB A newly isolated, purified, and characterized PQQ-dependent alc.

dehydrogenase (a bacterial membrane-bound protein) was recently found to display a surprisingly large linear range and high selectivity towards ethanol when integrated into a conducting polymer network on a platinum electrode. These findings motivated us to study the enzyme when simply immobilized onto carbonaceous surfaces in order to establish its characteristics and suitability for sensor development, the sensor design being based on a direct-electron transfer pathway. Graphite rods and screen-printed electrodes were modified in two different ways, and were operated both in FIA and batch mode. The obtained biosensor characteristics were highly dependent on the sensor architecture, the highest sensitivity (179 mA M<sup>-1</sup> cm<sup>-2</sup>) and lowest detection limit (1 .μ.M) being obtained for screen-printed electrodes used in a batch mode. A mechanism of the obsd. direct electron transfer between the enzyme's active center and the electrode is proposed.

- ST bioelectrocatalysis carbon electrode quinohemoprotein alc dehydrogenase  
 IT Conducting polymers  
     Enzyme electrodes  
     Gluconobacter  
     Immobilization, molecular  
     Screen printing  
         (bioelectrocatalysis at carbon electrodes modified with quinohemoprotein alc. dehydrogenase from Gluconobacter sp. 33)  
 IT Catalysis  
     (electrocatalysis; bioelectrocatalysis at carbon electrodes modified with quinohemoprotein alc. dehydrogenase from Gluconobacter sp. 33)  
 IT 64-17-5, Ethanol, analysis  
     RL: ANT (Analyte); ANST (Analytical study)  
         (bioelectrocatalysis at carbon electrodes modified with quinohemoprotein alc. dehydrogenase from Gluconobacter sp. 33)  
 IT 37205-43-9, E.C.1.1.99.8  
     RL: ARG (Analytical reagent use); PEP (Physical, engineering or chemical process); PYP (Physical process); ANST (Analytical study); PROC (Process); USES (Uses)  
         (bioelectrocatalysis at carbon electrodes modified with quinohemoprotein alc. dehydrogenase from Gluconobacter sp. 33)  
 IT 72909-34-3, PQQ  
     RL: ARU (Analytical role, unclassified); ANST (Analytical study)  
         (bioelectrocatalysis at carbon electrodes modified with quinohemoprotein alc. dehydrogenase from Gluconobacter sp. 33)  
 IT 30604-81-0, Polypyrrole  
     RL: ARU (Analytical role, unclassified); DEV (Device component use); ANST (Analytical study); USES (Uses)  
         (bioelectrocatalysis at carbon electrodes modified with quinohemoprotein alc. dehydrogenase from Gluconobacter sp. 33)  
 IT 7440-06-4, Platinum, uses 7440-44-0, Carbon,  
     uses 7440-57-5, Gold, uses 7782-42-5,  
     Graphite, uses  
     RL: DEV (Device component use); USES (Uses)  
         (bioelectrocatalysis at carbon electrodes modified with quinohemoprotein alc. dehydrogenase from Gluconobacter sp. 33)  
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IT 7440-06-4, Platinum, uses 7440-44-0, Carbon,  
 uses 7440-57-5, Gold, uses 7782-42-5,

Graphite, uses

RL: DEV (Device component use); USES (Uses)  
 (bioelectrocatalysis at carbon electrodes modified with  
 quinohemoprotein alc. dehydrogenase from Gluconobacter sp. 33)

RN 7440-06-4 HCAPLUS

CN Platinum (8CI, 9CI) (CA INDEX NAME)

Pt

RN 7440-44-0 HCAPLUS

CN Carbon (7CI, 8CI, 9CI) (CA INDEX NAME)

C

RN 7440-57-5 HCAPLUS

CN Gold (8CI, 9CI) (CA INDEX NAME)

Au

RN 7782-42-5 HCAPLUS  
CN Graphite (8CI, 9CI) (CA INDEX NAME)

C

L77 ANSWER 5 OF 33 HCAPLUS COPYRIGHT 2003 ACS  
AN 2001:482681 HCAPLUS  
DN 135:207620  
TI Detection of histamine and other biogenic amines using biosensors based on amine oxidase  
AU Niculescu, M.; Nistor, C.; Ruzgas, T.; Frebort, I.;  
Sebela, M.; Pec, P.; Csoregi, E.  
CS Department of Biotechnology, Lund University, Lund, S-22100, Swed.  
SO Inflammation Research (2001), 50(Suppl. 2), S146-S148  
CODEN: INREFB; ISSN: 1023-3830  
PB Birkhaeuser Verlag  
DT Journal  
LA English  
CC 9-2 (Biochemical Methods)  
Section cross-reference(s): 2  
AB The authors have developed two types of amine oxidase-based biosensors: a monoenzymic and a bienzymic one, the latter being based on co-immobilized amine oxidase (AO) and horseradish peroxidase (HRP). The design of the sensor is either based on a direct electron transfer or a mediated one. In general the bienzymic biosensors showed superior electrode characteristics than the monoenzymic ones, both for unmediated and mediated types (e.g., higher sensitivity, lower detection limit, larger dynamic range, etc.). However, the optimized monoenzymic biosensor surprisingly displayed very low sensitivity for putrescine in comparison with histamine. To clarify the obsd. difference in selectivity, the electron transfer mechanism of the two electrode types has to be elucidated. The present work targeted the interpretation of hypotheses explaining the possible electron transfer mechanism for the monoenzymic biosensor. When recording the current signals for various amines, the unmediated bienzymic (AO-HRP) biosensor followed the substrate specificity of the enzyme in soln., whereas the monoenzymic (AO) biosensor showed remarkably changed selectivity, responding mainly to histamine, cystamine and tyramine. The obtained results suggest that the electron transfer mechanism is a mixt. between a direct and an internally mediated one (via the electro-oxidn. of the formed product). However, the AO electrode is the first example when a copper AO can work anaerobically. An exptl. setup consisting of AO and AO-HRP electrodes can be thus used for the selective detection of histamine and diamines (putrescine and cadaverine) due to the difference in their selectivity pattern.  
ST biogenic amine biosensor amine oxidase;  
histamine biogenic amine biosensor amine oxidase  
IT Amines, analysis  
RL: ANT (Analyte); ANST (Analytical study)  
(biogenic; detection of histamine and other biogenic amines using biosensors based on amine oxidase)  
IT Biosensors  
(detection of histamine and other biogenic amines using biosensors based on amine oxidase)  
IT Electron transfer  
(detection of histamine and other biogenic amines using biosensors based on amine oxidase in

relation to electron transfer)

IT 51-45-6, Histamine, analysis 51-67-2, Tyramine 51-85-4, Cystamine  
 107-15-3, Ethylenediamine, analysis 110-60-1, Putrescine 124-20-9,  
 Spermidine 306-60-5, Agmatine 462-94-2, Cadaverine 40794-72-7  
 40930-37-8  
 RL: ANT (Analyte); ANST (Analytical study)  
 (detection of histamine and other biogenic amines using  
 biosensors based on amine oxidase)

IT 9059-11-4, Amine oxidase  
 RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)  
 (detection of histamine and other biogenic amines using  
 biosensors based on amine oxidase)

RE.CNT 8 THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE

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IT 9059-11-4, Amine oxidase

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)  
 (detection of histamine and other biogenic amines using  
 biosensors based on amine oxidase)

RN 9059-11-4 HCPLUS

CN Oxidase, amine (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L77 ANSWER 6 OF 33 HCPLUS COPYRIGHT 2003 ACS

AN 2001:451092 HCPLUS

DN 135:58124

TI Sensor element and its manufacturing method

IN Rui, Masao

PA Toto Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 11 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G01N027-327

CC 9-1 (Biochemical Methods)

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2001165892	A2	<20010622	JP 1999-347027	19991207
PRAI	JP 1999-347027		{19991207}-		

AB A sensor element used for an electrochem. measuring system is provided. In this sensor element, a responsive layer contg. a biocatalyst (e.g., glucose oxidase, uricase, glutamate oxidase, L-amino acid oxidase, D-amino acid oxidase, alc. oxidase, bilirubin oxidase, amine oxidase, cholesterol oxidase, choline oxidase, xanthine oxidase, pyruvate oxidase, lactate oxidase) capable of recognizing a target substance, and a selective permeable membrane for selectively prohibiting the permeation of a coexisting interfering substance causative of an undesirable electrochem. reaction are strongly held on the surface of its electricity collector by a phys. or chem. force. This capability is provided by processing at least a surface part of the electricity collector and turning it into a mixt. of metal (e.g, platinum, gold, silver, palladium, osmium, iridium,

carbon, nickel, iron, lead, copper), an inorg. substance (e.g., silicon, titanium, aluminum, tantalum) and an org. substance. Diagrams describing the **sensor** assembly are given.

ST electrochem **sensor** transducer enzyme **electrode** metal

IT Sensors

Transducers  
(electrochem.; **sensor** element and manufg. method)

IT Annealing  
(plasma; **sensor** element and manufg. method)

IT Annealing  
Coating process

Enzyme **electrodes**

Immobilization, biochemical

Membranes, nonbiological

Permeability  
(**sensor** element and manufg. method)

IT Enzymes, uses  
RL: ARG (Analytical reagent use); PEP (Physical, engineering or chemical process); ANST (Analytical study); PROC (Process); USES (Uses)  
(**sensor** element and manufg. method)

IT Metals, uses  
RL: DEV (Device component use); USES (Uses)  
(**sensor** element and manufg. method)

IT Electrochemical cells  
(transducers; **sensor** element and manufg. method)

IT 9035-73-8, Oxidase  
RL: ARG (Analytical reagent use); PEP (Physical, engineering or chemical process); ANST (Analytical study); PROC (Process); USES (Uses)  
(hydrogen peroxide-forming; **sensor** element and manufg.  
method)

IT 9000-88-8, D-Amino acid **oxidase** 9000-89-9, L-Amino acid  
**oxidase** 9001-37-0, Glucose **oxidase** 9001-96-1,  
Pyruvate **oxidase** 9002-12-4, Uricase 9002-17-9, Xanthine  
**oxidase** 9028-67-5, Choline **oxidase** 9028-72-2,  
Lactate **oxidase** 9028-76-6, Cholesterol **oxidase**  
9059-11-4, Amine **oxidase** 9073-63-6, Alcohol  
**oxidase** 39346-34-4, Glutamate **oxidase** 80619-01-8,  
Bilirubin **oxidase**  
RL: ARG (Analytical reagent use); PEP (Physical, engineering or chemical process); ANST (Analytical study); PROC (Process); USES (Uses)  
(**sensor** element and manufg. method)

IT 7429-90-5, Aluminum, uses 7439-88-5, Iridium, uses 7439-89-6, Iron,  
uses 7439-92-1, Lead, uses 7440-02-0, Nickel, uses 7440-04-2  
, Osmium, uses 7440-05-3, Palladium, uses  
7440-06-4, Platinum, uses 7440-21-3, Silicon, uses  
7440-22-4, Silver, uses 7440-25-7, Tantalum, uses  
7440-32-6, Titanium, uses 7440-44-0, Carbon, uses  
7440-50-8, Copper, uses 7440-57-5,  
Gold, uses  
RL: DEV (Device component use); USES (Uses)  
(**sensor** element and manufg. method)

IT 9059-11-4, Amine **oxidase**  
RL: ARG (Analytical reagent use); PEP (Physical, engineering or chemical process); ANST (Analytical study); PROC (Process); USES (Uses)  
(**sensor** element and manufg. method)

RN 9059-11-4 HCAPLUS

CN Oxidase, amine (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

IT 7440-04-2, Osmium, uses 7440-05-3,  
Palladium, uses 7440-06-4, Platinum, uses  
7440-22-4, Silver, uses 7440-44-0,  
Carbon, uses 7440-50-8, Copper, uses

7440-57-5, Gold, uses  
RL: DEV (Device component use); USES (Uses)  
(sensor element and manufg. method)

RN 7440-04-2 HCAPLUS  
CN Osmium (8CI, 9CI) (CA INDEX NAME)

Os

RN 7440-05-3 HCAPLUS  
CN Palladium (8CI, 9CI) (CA INDEX NAME)

Pd

RN 7440-06-4 HCAPLUS  
CN Platinum (8CI, 9CI) (CA INDEX NAME)

Pt

RN 7440-22-4 HCAPLUS  
CN Silver (8CI, 9CI) (CA INDEX NAME)

Ag

RN 7440-44-0 HCAPLUS  
CN Carbon (7CI, 8CI, 9CI) (CA INDEX NAME)

C

RN 7440-50-8 HCAPLUS  
CN Copper (7CI, 8CI, 9CI) (CA INDEX NAME)

Cu

RN 7440-57-5 HCAPLUS  
CN Gold (8CI, 9CI) (CA INDEX NAME)

Au

L77 ANSWER 7 OF 33 HCAPLUS COPYRIGHT 2003 ACS  
AN 2001:343486 HCAPLUS  
DN 135:2401  
TI Interference elimination in glutamate monitoring with chip integrated  
enzyme microreactors  
AU Collins, A.; Mikeladze, E.; Bengtsson, M.; Kokaia, M.; Laurell, T.;  
Csoregi, E.  
CS Department of Biotechnology, Lund University, Lund, S-22100, Swed.  
SO Electroanalysis (2001), 13(6), 425-431  
CODEN: ELANEU; ISSN: 1040-0397

PB Wiley-VCH Verlag GmbH

DT Journal

LA English

CC 9-1 (Biochemical Methods)

AB On-chip enzyme reactors are often used in medical/pharmaceutical anal. due to their inherent advantages, such as high sample throughput, low reagent consumption, stability, reproducibility and low cost. The present work describes a different application of such microreactors, namely, elimination of interfering ascorbate signals in glutamate monitoring using ascorbate **oxidase** modified silicon chip microreactors of different sizes (5.3 and 0.95 .mu.L). Glutamate was monitored with a previously developed **redox hydrogel** integrated bienzyme electrode, based on coupled glutamate **oxidase** and **horseradish peroxidase**, inserted in a miniaturized flow cell operated at - 50 mV (vs. Ag/AgCl). The developed online anal. system was characterized with regard to diln. effects, detection limit, response time and interference ability using model solns. and real samples. Off-line in vivo glutamate measurements could be made by injecting rat brain microdialyzate samples collected before and after KCl stimulation without any interference of ascorbate. Within the studied flow rate range (2-25 .mu.L/min), 1 mM and 200 .mu.M ascorbate could be totally eliminated using the larger and the smaller microreactor, resp.

ST glutamate biochip integrated enzyme microreactor interference elimination  
IT Bioreactors

#### Biosensors

Enzyme electrodes

Immobilization, biochemical

Interference

(glutamate monitoring with chip integrated enzyme microreactors)

IT 56-86-0, Glutamic acid, analysis

RL: ANT (Analyte); ANST (Analytical study)

(glutamate monitoring with chip integrated enzyme microreactors)

IT 39346-34-4, Glutamate **oxidase**

RL: ARG (Analytical reagent use); PEP (Physical, engineering or chemical process); ANST (Analytical study); PROC (Process); USES (Uses)

(glutamate monitoring with chip integrated enzyme microreactors)

IT 50-81-7, Ascorbic acid, analysis

RL: ARU (Analytical role, unclassified); ANST (Analytical study)

(glutamate monitoring with chip integrated enzyme microreactors)

IT 7440-21-3, Silicon, uses

RL: DEV (Device component use); USES (Uses)

(glutamate monitoring with chip integrated enzyme microreactors)

IT 9003-99-0, **Peroxidase**

RL: ARG (Analytical reagent use); PEP (Physical, engineering or chemical process); ANST (Analytical study); PROC (Process); USES (Uses)

(**horseradish**; glutamate monitoring with chip integrated enzyme microreactors)

RE.CNT 54 THERE ARE 54 CITED REFERENCES AVAILABLE FOR THIS RECORD

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IT 9003-99-0, Peroxidase

RL: ARG (Analytical reagent use); PEP (Physical, engineering or chemical process); ANST (Analytical study); PROC (Process); USES (Uses)  
 (horseradish; glutamate monitoring with chip integrated enzyme microreactors)

RN 9003-99-0 HCPLUS

CN Peroxidase (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L77 ANSWER 8 OF 33 HCPLUS COPYRIGHT 2003 ACS  
 AN 2001:88137 HCPLUS  
 DN 134:216600  
 TI An oxygen-independent ethanol sensor based on quinohemoprotein alcohol dehydrogenase covalently bound to a functionalized polypyrrole film  
 AU Ramanavicius, A.; Habermuller, K.; Razumiene, J.; Meskys, R.; Marcinkeviciene, L.; Bachmatova, I.; Csoregi, E.; Laurinavicius, V.; Schuhmann, W.  
 CS Laboratory of Bioanalysis, Institute of Biochemistry, Vilnius, 2600, Lithuania  
 SO Progress in Colloid & Polymer Science (2000), 116(Surface and Colloid Science), 143-148  
 CODEN: PCPSD7; ISSN: 0340-255X

PB Springer  
 DT Journal  
 LA English  
 CC 80-2 (Organic Analytical Chemistry)  
 Section cross-reference(s): 9, 72  
 AB The characteristics of a phenazine methosulfate mediated alc.  
**biosensor** based on a newly isolated quinohemoprotein alc.  
 dehydrogenase are described. The enzyme was covalently linked at a  
 functionalized polypyrrole film which had been electrochem. deposited on  
 the surface of a ~~platinum-black electrode~~. The  
**biosensor** architecture developed was characterized with regard to  
 sensitivity, selectivity, and long-term operational stability. Owing to  
 the inherent properties of the new enzyme the related **biosensors**  
 are oxygen-independent and exhibit improved selectivity to ethanol in  
 contrast to alc. **biosensors** based on alc. oxidase or  
 on cationic NAD dependent alc. dehydrogenase.  
 ST ethanol **biosensor** alc dehydrogenase functionalized polypyrrole  
 IT Enzyme electrodes  
     (amperometric; oxygen-independent ethanol sensor based on  
     quinohemoprotein alc. dehydrogenase covalently bound to a  
     functionalized polypyrrole film)  
 IT Biosensors  
     Cyclic voltammetry  
     (oxygen-independent ethanol sensor based on quinohemoprotein alc.  
     dehydrogenase covalently bound to a functionalized polypyrrole film)  
 IT Alcohols, analysis  
     RL: ARU (Analytical role, unclassified); ANST (Analytical study)  
     (response of oxygen-independent ethanol sensor based on  
     quinohemoprotein alc. dehydrogenase covalently bound to a  
     functionalized polypyrrole film to)  
 IT 109-97-7, Pyrrole 3251-23-8  
     RL: ARU (Analytical role, unclassified); RCT (Reactant); ANST (Analytical  
     study); RACT (Reactant or reagent)  
     (in prepn. of oxygen-independent ethanol sensor based on  
     quinohemoprotein alc. dehydrogenase covalently bound to a  
     functionalized polypyrrole film)  
 IT 64-17-5, Ethanol, analysis  
     RL: ANT (Analyte); ANST (Analytical study)  
     (oxygen-independent ethanol sensor based on quinohemoprotein alc.  
     dehydrogenase covalently bound to a functionalized polypyrrole film)  
 IT 9031-72-5, Alcohol dehydrogenase 30604-81-0, Polypyrrole  
     RL: ARU (Analytical role, unclassified); DEV (Device component use); ANST  
     (Analytical study); USES (Uses)  
     (oxygen-independent ethanol sensor based on quinohemoprotein alc.  
     dehydrogenase covalently bound to a functionalized polypyrrole film)  
 IT 7440-06-4, Platinum-black, analysis  
     RL: ARU (Analytical role, unclassified); DEV (Device component use); ANST  
     (Analytical study); USES (Uses)  
     (platinum-black electrode; oxygen-independent ethanol sensor  
     based on quinohemoprotein alc. dehydrogenase covalently bound to a  
     functionalized polypyrrole film)  
 IT 67-56-1, Methanol, analysis 71-23-8, 1-Propanol, analysis 71-36-3,  
 1-Butanol, analysis 78-83-1, Isobutanol, analysis  
     RL: ARU (Analytical role, unclassified); ANST (Analytical study)  
     (response of oxygen-independent ethanol sensor based on  
     quinohemoprotein alc. dehydrogenase covalently bound to a  
     functionalized polypyrrole film to)  
 RE.CNT 34 THERE ARE 34 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 RE  
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IT 7440-06-4, Platinum-black, analysis

RL: ARU (Analytical role, unclassified); DEV (Device component use); ANST (Analytical study); USES (Uses)  
 (platinum-black electrode; oxygen-independent ethanol sensor based on quinohemoprotein alc. dehydrogenase covalently bound to a functionalized polypyrrole film)

RN 7440-06-4 HCPLUS

CN Platinum (8CI, 9CI) (CA INDEX NAME)

Pt

- L77 ANSWER 9 OF 33 HCPLUS COPYRIGHT 2003 ACS  
 AN 2001:88010 HCPLUS  
 DN 134:219097  
 TI Hydrogen peroxide sensitive biosensor based on plant peroxidases entrapped in Os-modified polypyrrole films  
 AU Gaspar, Szilveszter; Habermuller, Katja; Csoregi, Elisabeth; Schuhmann, Wolfgang  
 CS Department of Biotechnology, University of Lund, Lund, S-22100, Swed.  
 SO Sensors and Actuators, B: Chemical (2001), B72(1), 63-68  
 CODEN: SABCEB; ISSN: 0925-4005  
 PB Elsevier Science S.A.  
 DT Journal  
 LA English  
 CC 9-1 (Biochemical Methods)  
 AB An amperometric hydrogen peroxide biosensor was designed based on horseradish and tobacco peroxidase entrapped into a conducting redox-polymer immobilized on either glassy-carbon or platinum electrodes. A versatile one-step

- immobilization method was carried out based on the electrochem. polymn. of a pyrrole monomer functionalized with an Os-complex. Cyclic voltammetry and const. potential amperometry performed with the different peroxidases in soln. or entrapped within the conducting redox-polymer film suggests that the redox center within the active site of horseradish peroxidase exhibits a better accessibility for the either free-diffusing or polymer-bound Os-complexes than that of tobacco peroxidase. Therefore, the obtained sensitivities for the redn. of H<sub>2</sub>O<sub>2</sub> are significantly higher for the HRP-based sensors as compared with the tobacco peroxidase-based ones. The direct redn. of H<sub>2</sub>O<sub>2</sub> on the polymer backbone was identified as a side reaction even though the bioelectroredn. through horseradish peroxidase is a much more efficient reaction pathway.
- ST    hydrogen peroxide biosensor plant peroxidase  
      Os polypyrrole electrode
- IT    Amperometry  
      Conducting polymers  
      Cyclic voltammetry  
            Horseradish (*Armoracia lapathifolia*)  
      Immobilization, biochemical  
            Tobacco  
                (hydrogen peroxide sensitive biosensor based on plant peroxidases entrapped in Os-modified polypyrrole films)
- IT    7722-84-1, Hydrogen peroxide, analysis  
      RL: ANT (Analyte); ANST (Analytical study)  
            (hydrogen peroxide sensitive biosensor based on plant peroxidases entrapped in Os-modified polypyrrole films)
- IT    9003-99-0, Peroxidase  
      RL: ARG (Analytical reagent use); DEV (Device component use); ANST (Analytical study); USES (Uses)  
            (hydrogen peroxide sensitive biosensor based on plant peroxidases entrapped in Os-modified polypyrrole films)
- IT    7440-06-4, Platinum, uses 7440-44-0, Carbon,  
      uses  
      RL: DEV (Device component use); USES (Uses)  
            (hydrogen peroxide sensitive biosensor based on plant peroxidases entrapped in Os-modified polypyrrole films)
- IT    329353-89-1P  
      RL: DEV (Device component use); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)  
            (hydrogen peroxide sensitive biosensor based on plant peroxidases entrapped in Os-modified polypyrrole films)
- RE.CNT 38    THERE ARE 38 CITED REFERENCES AVAILABLE FOR THIS RECORD
- RE
- (1) Bartlett, P; J Electroanal Chem 1993, V362, P1 HCPLUS
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IT 9003-99-0, Peroxidase

RL: ARG (Analytical reagent use); DEV (Device component use); ANST (Analytical study); USES (Uses)  
 (hydrogen peroxide sensitive biosensor based on plant peroxidases entrapped in Os-modified polypyrrole films)

RN 9003-99-0 HCPLUS

CN Peroxidase (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

IT 7440-06-4, Platinum, uses 7440-44-0, Carbon,  
 uses

RL: DEV (Device component use); USES (Uses)  
 (hydrogen peroxide sensitive biosensor based on plant peroxidases entrapped in Os-modified polypyrrole films)

RN 7440-06-4 HCPLUS

CN Platinum (8CI, 9CI) (CA INDEX NAME)

Pt

RN 7440-44-0 HCPLUS  
 CN Carbon (7CI, 8CI, 9CI) (CA INDEX NAME)

C

L77 ANSWER 10 OF 33 HCPLUS COPYRIGHT 2003 ACS  
 AN 2001:31711 HCPLUS  
 DN 134:85306  
 TI Biosensor for determination of freshness biomarkers in food and beverage

IN **Csoregi, Elisabeth; Niculescu, Mihaela; Frebort, Ivo**

PA **Forskarpatent i Syd AB, Swed.**

SO **PCT Int. Appl., 22 pp.**

**CODEN: PIXXD2**

DT **Patent**

LA **English**

IC **ICM G01N**

CC **17-1 (Food and Feed Chemistry)**

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2001002827	A2	20010111	WO 2000-SE1449	20000706 <--
	WO 2001002827	A3	20010628		
		W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM		
		RW:	GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG		
	AU 2000060439	A5	20010122	AU 2000-60439	20000706 <--
	EP 1198588	A2	20020424	EP 2000-946725	20000706 <--
		R:	AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL		
PRAI	SE 1999-2608	A	19990706	<--	
	WO 2000-SE1449	W	20000706	<--	

AB The present invention relates to a **biosensor** for the detection and/or the detn. of freshness biomarkers in foods and beverages, comprising an **electrode** and a mono-enzyme system, such as an **amine oxidase**, or a bi-enzyme system of an **amine oxidase** and a **peroxidase**.

ST **biosensor electrode amine oxidase peroxidase food freshness**

IT **Electrodes**

(**bioelectrodes; biosensor for detn. of freshness biomarkers in foods and beverages**)

IT **Amines, analysis**

RL: **ANT (Analyte); ANST (Analytical study)**

(**biogenic; biosensor for detn. of freshness biomarkers in foods and beverages**)

IT **Beverages**

**Blood analysis**

**Body fluid**

**Diagnosis**

**Dialysis fluids**

**Disease, animal**

**Electrodes**

**Fish**

**Food analysis**

**Meat**

**Saliva**

**Sweat**

**Urine analysis**

(**biosensor for detn. of freshness biomarkers in foods and beverages**)

IT **Enzymes, uses**

RL: **ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)**

(**biosensor for detn. of freshness biomarkers in foods and beverages**)

IT **Carbon fibers, uses**

RL: DEV (Device component use); USES (Uses)  
     (biosensor for detn. of freshness biomarkers in foods and  
     beverages)

IT Paste electrodes  
     (carbon; biosensor for detn. of freshness  
     biomarkers in foods and beverages)

IT Polymers, uses  
     Salts, uses  
     RL: DEV (Device component use); USES (Uses)  
         (conducting; biosensor for detn. of freshness biomarkers in  
         foods and beverages)

IT 51-45-6, Histamine, analysis 110-60-1, Putrescine  
     RL: ANT (Analyte); ANST (Analytical study)  
         (biosensor for detn. of freshness biomarkers in foods and  
         beverages)

IT 9003-99-0, Peroxidase 9059-11-4, Amine  
     oxidase 25232-42-2D, Poly(1-vinylimidazole),  
     complexes with Os(4,4'-dimethyl-bipyridine  
     )+/2+ and poly(ethylene glycol)  
     diglycidyl-ether  
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)  
         (biosensor for detn. of freshness biomarkers in foods and  
         beverages)

IT 7440-05-3, Palladium, uses 7440-06-4,  
     Platinum, uses 7440-22-4, Silver, uses  
     7440-44-0, Carbon, uses 7440-57-5,  
     Gold, uses 7782-42-5, Graphite, uses  
     RL: DEV (Device component use); USES (Uses)  
         (biosensor for detn. of freshness biomarkers in foods and  
         beverages)

IT 9003-99-0, Peroxidase 9059-11-4, Amine  
     oxidase 25232-42-2D, Poly(1-vinylimidazole),  
     complexes with Os(4,4'-dimethyl-bipyridine  
     )+/2+ and poly(ethylene glycol)  
     diglycidyl-ether  
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)  
         (biosensor for detn. of freshness biomarkers in foods and  
         beverages)

RN 9003-99-0 HCAPLUS  
 CN Peroxidase (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

RN 9059-11-4 HCAPLUS  
 CN Oxidase, amine (9CI) (CA INDEX NAME)

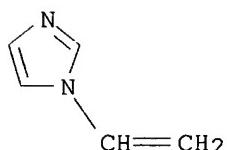
\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

RN 25232-42-2 HCAPLUS  
 CN 1H-Imidazole, 1-ethenyl-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 1072-63-5

CMF C5 H6 N2



IT 7440-05-3, Palladium, uses 7440-06-4,  
 Platinum, uses 7440-22-4, Silver, uses  
 7440-44-0, Carbon, uses 7440-57-5,  
 Gold, uses 7782-42-5, Graphite, uses  
 RL: DEV (Device component use); USES (Uses)  
 (biosensor for detn. of freshness biomarkers in foods and  
 beverages)

RN 7440-05-3 HCPLUS  
 CN Palladium (8CI, 9CI) (CA INDEX NAME)

## Pd

RN 7440-06-4 HCPLUS  
 CN Platinum (8CI, 9CI) (CA INDEX NAME)

## Pt

RN 7440-22-4 HCPLUS  
 CN Silver (8CI, 9CI) (CA INDEX NAME)

## Ag

RN 7440-44-0 HCPLUS  
 CN Carbon (7CI, 8CI, 9CI) (CA INDEX NAME)

## C

RN 7440-57-5 HCPLUS  
 CN Gold (8CI, 9CI) (CA INDEX NAME)

## Au

RN 7782-42-5 HCPLUS  
 CN Graphite (8CI, 9CI) (CA INDEX NAME)

## C

L77 ANSWER 11 OF 33 / HCPLUS COPYRIGHT 2003 ACS  
 AN 2000:870552 HCPLUS  
 DN 134:159625  
 TI Biosensors based on novel plant peroxidases: a  
 comparative study  
 AU Gaspar, S.; Popescu, I. C.; Gazaryan, I. G.; Gerardo Bautista, A.;  
 Sakharov, I. Y.; Mattiasson, B.; Csoregi, E.  
 CS Department of Biotechnology, Lund University, Lund, SE-22100, Swed.  
 SO Electrochimica Acta (2000), 46(2-3), 255-264  
 CODEN: ELCAAV; ISSN: 0013-4686  
 PB Elsevier Science Ltd.  
 DT Journal  
 LA English

- CC 9-1 (Biochemical Methods)  
 Section cross-reference(s): 7
- AB Amperometric biosensors for hydrogen peroxide detection have been constructed using horseradish peroxidase (HRP) and two newly purified peroxidases extd. from tobacco (TOP) and sweet potato (SPP). The peroxidases were cross-linked to a redox polymer [poly(vinylimidazole) complexed with Os(4,4'dimethylbipyridine)2Cl2] using poly(ethylene glycol) diglycidyl ether as the crosslinker. A comparative study with regard to their bioelectrochem. characteristics showed that, irresp. of peroxidase, the biosensors sensitivity was strongly influenced by hydrogel compn., curing procedure, film thickness and applied potential. The electrostatic interaction between the cationic redox polymer and the neg. charged peroxidases (TOP and SPP) enhanced the hydrogen peroxide signal. When operated in a FI system, the optimized SPP biosensor (48% redox polymer, 23% cross-linker and 29% enzyme, wt./wt. %) displayed the highest sensitivity for H<sub>2</sub>O<sub>2</sub> (3.2 A M-1cm-2), a linear range up to 220 .mu.M, a detection limit of 25 nM (calcd. as 2S/N) and a response time of about 2 min.
- ST peroxidase osmium redox polymer hydrogen peroxide detn
- IT Enzyme electrodes  
 (amperometric; hydrogen peroxide detection using plant peroxidases coupled to osmium redox polymer )
- IT Immobilization, biochemical  
 (enzyme; hydrogen peroxide detection using plant peroxidases coupled to osmium redox polymer)
- IT Electron transfer  
 Enzyme kinetics  
 (hydrogen peroxide detection using plant peroxidases coupled to osmium redox polymer)
- IT Hydrogels  
 (redox; hydrogen peroxide detection using plant peroxidases coupled to osmium redox polymer)
- IT 9003-99-0, Peroxidase  
 RL: ARG (Analytical reagent use); PEP (Physical, engineering or chemical process); ANST (Analytical study); PROC (Process); USES (Uses)  
 (horseradish; hydrogen peroxide detection using plant peroxidases coupled to osmium redox polymer )
- IT 7722-84-1, Hydrogen peroxide, analysis  
 RL: ANT (Analyte); ANST (Analytical study)  
 (hydrogen peroxide detection using plant peroxidases coupled to osmium redox polymer)
- IT 15320-22-6D, complexes with polyvinylimidazole and epoxy resins 25232-42-2D, Osmium bipyridine chloride epoxy resin complexes 26403-72-5D, Polyethylene glycol diglycidyl ether, Osmium bipyridine chloride complexes contg. polyvinylimidazole  
 RL: ARU (Analytical role, unclassified); ANST (Analytical study)  
 (hydrogen peroxide detection using plant peroxidases coupled to osmium redox polymer)
- RE.CNT 49 THERE ARE 49 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 RE  
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IT 9003-99-0, Peroxidase

RL: ARG (Analytical reagent use); PEP (Physical, engineering or chemical process); ANST (Analytical study); PROC (Process); USES (Uses)  
 (horseradish; hydrogen peroxide detection using plant peroxidases coupled to osmium redox polymer  
 )

RN 9003-99-0 HCPLUS

CN Peroxidase (9CI) (CA INDEX NAME)

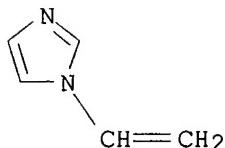
\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

IT 25232-42-2D, Osmium bipyridine chloride epoxy resin complexes 26403-72-5D, Polyethylene glycol diglycidyl ether, Osmium bipyridine chloride complexes contg. polyvinylimidazole  
 RL: ARU (Analytical role, unclassified); ANST (Analytical study)  
 (hydrogen peroxide detection using plant peroxidases coupled to osmium redox polymer)

RN 25232-42-2 HCPLUS

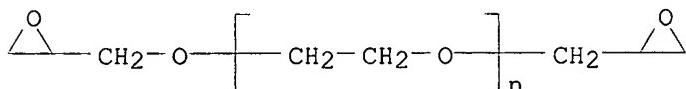
CN 1H-Imidazole, 1-ethenyl-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 1072-63-5  
CMF C5 H6 N2

RN 26403-72-5 HCPLUS

CN Poly(oxy-1,2-ethanediyl), .alpha.- (oxiranylmethyl)-.omega.- (oxiranylmethoxy)- (9CI) (CA INDEX NAME)



L77 ANSWER 12 OF 33 HCPLUS COPYRIGHT 2003 ACS

AN 2000:853721 HCPLUS

DN 134:127954

TI Biosensors based on novel peroxidases with improved properties in direct and mediated electron transfer

AU Lindgren, A.; Ruzgas, T.; Gorton, L.; Csoregi, E.; Bautista Ardila, G.; Sakharov, I. Y.; Gazaryan, I. G.

CS Department of Analytical Chemistry, Lund University, Lund, SE-22100, Swed.

SO Biosensors &amp; Bioelectronics (2000), 15(9-10), 491-497

CODEN: BBIOE4; ISSN: 0956-5663

PB Elsevier Science S.A.

DT Journal

LA English

CC 9-1 (Biochemical Methods)

Section cross-reference(s): 7

AB Native horseradish peroxidase (HRP) on graphite has revealed apprxeq. 50% of the active enzyme mols. to be in direct electron transfer (ET) contact with the electrode surface.

Some novel plant peroxidases from tobacco, peanut and sweet potato were kinetically characterized on graphite in order to find promising candidates for

biosensor applications and to understand the nature of the direct ET in the case of plant peroxidases. From measurements of the mediated and mediatorless currents of hydrogen peroxide redn. at the

peroxidase-modified rotating disk electrodes (RDE), it was concluded that the fraction of enzyme mols. in direct ET varies substantially for the different plant peroxidases. It was obsd.

that the anionic peroxidases (from sweet

potato and tobacco) demonstrated a higher percentage of mols. in direct ET than the cationic ones (HRP and peanut peroxidase). The peroxidases with a high degree of

glycosylation demonstrated a lower percentage of mols. in direct ET. It could, thus, be concluded that glycosylation of the peroxidases hinders direct ET and that a net neg. charge on the peroxidase (low pI value) is beneficial for direct ET. Esp. noticeable are the values obtained for sweet potato peroxidase

(SPP), revealing both a high percentage in direct ET and a high rate const. of direct ET. The peroxidase electrodes were used for detn. of hydrogen peroxide in RDE mode (mediatorless). SPP gave the lowest detection limit (40 nM) followed by HRP and peanut peroxidase.

- ST biosensor electron transfer plant peroxidase  
 IT Enzyme kinetics  
     (of inhibition; plant peroxidases as alternatives to HRP in peroxidase based biosensors)  
 IT Electron transfer  
     Enzyme electrodes  
         (plant peroxidases as alternatives to HRP in peroxidase based biosensors)  
 IT 9003-99-0, Peroxidase  
     RL: ARU (Analytical role, unclassified); PEP (Physical, engineering or chemical process); ANST (Analytical study); PROC (Process)  
         (plant; plant peroxidases as alternatives to HRP in peroxidase based biosensors)

RE.CNT 29 THERE ARE 29 CITED REFERENCES AVAILABLE FOR THIS RECORD

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- IT 9003-99-0, Peroxidase  
     RL: ARU (Analytical role, unclassified); PEP (Physical, engineering or chemical process); ANST (Analytical study); PROC (Process)  
         (plant; plant peroxidases as alternatives to HRP in peroxidase based biosensors)

RN 9003-99-0 HCPLUS  
 CN Peroxidase (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L77 ANSWER 13 OF 33 HCPLUS COPYRIGHT 2003 ACS  
 AN 2000:805378 HCPLUS  
 DN 134:97280  
 TI Electrooxidation Mechanism of Biogenic Amines at Amine

**Oxidase Modified Graphite Electrode**

AU Niculescu, Mihaela; Ruzgas, Tautgirdas; Nistor, Catalin;  
 Frebort, Ivo; Sebela, Marek; Pec, Pavel; Csoeregi, Elisabeth

CS Department of Biotechnology, Lund University, Lund, S-22100, Swed.

SO Analytical Chemistry (2000); 72(24), 5988-5993  
 CODEN: ANCHAM; ISSN: 0003-2700

PB American Chemical Society

DT Journal

LA English

CC 9-1 (Biochemical Methods)  
 Section cross-reference(s): 7, 72

AB Amine oxidase (AO, EC. 1.4.3.6) was previously shown to be a very efficient biol. recognition element of amperometric biosensors for monitoring biogenic amines. The enzyme was effectively working in both mono- and bienzyme electrode designs, based on either a direct or a mediated electron-transfer pathway. This work focuses on the elucidation of the electron-transfer mechanism of the monoenzymic unmediated AO-modified biosensor. The obsd. unmediated catalytic currents were assumed to be caused by (i) a direct electron-transfer process, (ii) the electrooxidn. of the formed product, or (iii) their combination. Expts. supporting these assumptions are discussed in detail.

ST electrooxidn mechanism biogenic **amine oxidase graphite electrode**

IT Amines, analysis  
 RL: ANT (Analyte); RCT (Reactant); ANST (Analytical study); RACT (Reactant or reagent)  
 (biogenic; electrooxidn. mechanism of biogenic amines at **amine oxidase modified graphite electrode**)

IT Biosensors  
 Chronoamperometry  
 Electrode reaction kinetics  
 Electron transfer  
 Enzyme electrodes  
 Enzyme kinetics  
 Oxidation, electrochemical  
 (electrooxidn. mechanism of biogenic amines at **amine oxidase modified graphite electrode**)

IT 147-84-2, reactions  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (copper complexation agent; electrooxidn. mechanism of biogenic amines at **amine oxidase modified graphite electrode**)

IT 56-17-7, Cystamine dihydrochloride 56-92-8, Histamine dihydrochloride  
 60-19-5, Tyramine hydrochloride 333-93-7, Putrescine dihydrochloride  
 1476-39-7, Cadaverine dihydrochloride 2482-00-0, Agmatine sulfate  
 49721-50-8, Spermidine phosphate  
 RL: ANT (Analyte); RCT (Reactant); ANST (Analytical study); RACT (Reactant or reagent)  
 (electrooxidn. mechanism of biogenic amines at **amine oxidase modified graphite electrode**)

IT 9003-99-0, Peroxidase  
 RL: ARG (Analytical reagent use); CAT (Catalyst use); DEV (Device component use); ANST (Analytical study); USES (Uses)  
 (electrooxidn. mechanism of biogenic amines at **amine oxidase modified graphite electrode**)

IT 9059-11-4, Amine oxidase  
 RL: ARG (Analytical reagent use); CAT (Catalyst use); DEV (Device component use); RCT (Reactant); ANST (Analytical study); RACT (Reactant or reagent); USES (Uses)  
 (electrooxidn. mechanism of biogenic amines at **amine oxidase modified graphite electrode**)

IT 9059-11-4DP, Amine oxidase, copper

-free

RL: ARG (Analytical reagent use); CAT (Catalyst use); DEV (Device component use); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses)  
 (electrooxidn. mechanism of biogenic amines at amine oxidase modified graphite electrode)

IT 7782-42-5, Graphite, uses

RL: DEV (Device component use); USES (Uses)  
 (electrooxidn. mechanism of biogenic amines at amine oxidase modified graphite electrode)

IT 7440-50-8, Copper, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)  
 (electrooxidn. mechanism of biogenic amines at amine oxidase modified graphite electrode)

RE.CNT 40 THERE ARE 40 CITED REFERENCES AVAILABLE FOR THIS RECORD

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IT 9003-99-0, Peroxidase

RL: ARG (Analytical reagent use); CAT (Catalyst use); DEV (Device component use); ANST (Analytical study); USES (Uses)  
 (electrooxidn. mechanism of biogenic amines at amine oxidase modified graphite electrode)

RN 9003-99-0 HCPLUS

CN Peroxidase (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

IT 9059-11-4, Amine oxidase

RL: ARG (Analytical reagent use); CAT (Catalyst use); DEV (Device component use); RCT (Reactant); ANST (Analytical study); RACT (Reactant or reagent); USES (Uses)

(electrooxidn. mechanism of biogenic amines at **amine oxidase modified graphite electrode**)

RN 9059-11-4 HCPLUS

CN Oxidase, amine (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

IT 9059-11-4DP, Amine oxidase, copper

-free

RL: ARG (Analytical reagent use); CAT (Catalyst use); DEV (Device component use); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses)

(electrooxidn. mechanism of biogenic amines at **amine oxidase modified graphite electrode**)

RN 9059-11-4 HCPLUS

CN Oxidase, amine (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

IT 7782-42-5, Graphite, uses

RL: DEV (Device component use); USES (Uses)

(electrooxidn. mechanism of biogenic amines at **amine oxidase modified graphite electrode**)

RN 7782-42-5 HCPLUS

CN Graphite (8CI, 9CI) (CA INDEX NAME)

C

IT 7440-50-8, Copper, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(electrooxidn. mechanism of biogenic amines at **amine oxidase modified graphite electrode**)

RN 7440-50-8 HCPLUS

CN Copper (7CI, 8CI, 9CI) (CA INDEX NAME)

Cu

L77 ANSWER 14 OF 33 HCPLUS COPYRIGHT 2003 ACS

AN 2000:749032 HCPLUS

DN 133:307286

TI Biosensor using plasma-polymerized membrane

IN Muguruma, Hitoshi; Hiratsuka, Akinori; Karube, Masao

PA Sentan Kagaku Gijutsu Incubation Center K. K., Japan

SO Jpn. Kokai Tokkyo Koho, 11 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G01N027-327

CC 9-1 (Biochemical Methods)

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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JP 2000298111	A2	20001024	JP 1999-107691	19990415

WO 2000063685 A1 20001026 WO 2000-JP2417 20000413  
 W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR,  
 CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU,  
 ID, IL, IN, IS, KE, KG, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA,  
 MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI,  
 SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM,  
 AZ, BY, KG, KZ, MD, RU, TJ, TM  
 RW: GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE,  
 DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF,  
 CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG

EP 1182450 A1 20020227 EP 2000-915512 20000413  
 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,  
 IE, SI, LT, LV, FI, RO

PRAI JP 1999-107691 A 19990415  
 WO 2000-JP2417 W 20000413

AB A highly functional biosensor with a novel structure is conveniently constructed using a plasma-polymd. membrane. The biosensor is constituted with a plasma-polymd. membrane contg. functional groups, a catalytically active substance (e.g., enzyme) immobilized on the plasma-polymd. membrane using a crosslinking reagent, and a metal electrode pattern in contact with a sample through the plasma-polymd. membrane. The influence by interfering compds. is eliminated due to the hydrogen peroxide-selective permeability of the membrane. The sensor can be applied in a wide range of areas in combination with micromachine technique. A diagram describing the sensor assembly is given.

ST biosensor plasma polymn membrane enzyme electrode

IT Amide group

Amino group

Carbonyl group

Carboxyl group

Crosslinking agents

Enzyme electrodes

Epoxy group

Formyl group

Functional groups

Glucose sensors

Hydroxyl group

Immobilization, biochemical

Membrane electrodes

Membranes, nonbiological

Micromachines

Permeability

Sulfhydryl group

(biosensor using plasma-polymd. membrane)

IT Enzymes, uses

RL: ARG (Analytical reagent use); DEV (Device component use); PEP (Physical, engineering or chemical process); ANST (Analytical study); PROC (Process); USES (Uses)

(biosensor using plasma-polymd. membrane)

IT Metals, uses

RL: DEV (Device component use); USES (Uses)

(biosensor using plasma-polymd. membrane)

IT Halogens

RL: NUU (Other use, unclassified); USES (Uses)

(biosensor using plasma-polymd. membrane)

IT Monomers

RL: NUU (Other use, unclassified); USES (Uses)

(biosensor using plasma-polymd. membrane)

IT Noble gases, uses

RL: NUU (Other use, unclassified); USES (Uses)

(biosensor using plasma-polymd. membrane)

IT Sensors  
 (electrochem.; biosensor using plasma-polymd.  
 membrane)

IT Functional groups  
 (imino group; biosensor using plasma-polymd.  
 membrane)

IT Functional groups  
 (isocyanato group; biosensor using plasma-polymd.  
 membrane)

IT Polymerization  
 (plasma; biosensor using plasma-polymd. membrane)

IT Functional groups  
 (vinyl group; biosensor using plasma-polymd.  
 membrane)

IT 7722-84-1, Hydrogen peroxide, analysis  
 RL: ANT (Analyte); PEP (Physical, engineering or chemical process); ANST  
 (Analytical study); PROC (Process)  
 (biosensor using plasma-polymd. membrane)

IT 9000-88-8, D-Amino acid oxidase 9000-89-9, L-Amino acid  
 oxidase 9001-37-0, Glucose oxidase 9001-46-1,  
 Glutamate dehydrogenase 9001-96-1, Pyruvate oxidase  
 9028-14-2, Glycerol dehydrogenase 9028-53-9, Glucose dehydrogenase  
 9028-67-5, Choline oxidase 9028-76-6, Cholesterol  
 oxidase 9028-79-9, Galactose oxidase 9028-86-8,  
 Aldehyde dehydrogenase 9031-72-5, Alcohol dehydrogenase 9035-73-8,  
 Oxidase 9035-82-9, Dehydrogenase 9059-11-4,  
 Amine oxidase 67775-34-2, Cholesterol dehydrogenase  
 135622-84-3, Fructose dehydrogenase 220983-94-8, Sorbitol dehydrogenase  
 RL: ARG (Analytical reagent use); DEV (Device component use); PEP  
 (Physical, engineering or chemical process); ANST (Analytical study); PROC  
 (Process); USES (Uses)  
 (biosensor using plasma-polymd. membrane)

IT 50-81-7, Ascorbic acid, analysis 51-61-6, Dopamine, analysis 57-13-6,  
 Urea, analysis 103-90-2  
 RL: ARU (Analytical role, unclassified); ANST (Analytical study)  
 (biosensor using plasma-polymd. membrane)

IT 7440-06-4, Platinum, uses  
 RL: DEV (Device component use); USES (Uses)  
 (biosensor using plasma-polymd. membrane)

IT 302-01-2, Hydrazine, uses 1333-74-0, Hydrogen, uses 7664-41-7,  
 Ammonia, uses 7727-37-9, Nitrogen, uses 7732-18-5, Water, uses  
 7782-44-7, Oxygen, uses 7783-06-4, Hydrogen sulfide, uses 13465-07-1,  
 Hydrogen disulfide  
 RL: NUU (Other use, unclassified); USES (Uses)  
 (biosensor using plasma-polymd. membrane)

IT 75-05-8, Acetonitrile, reactions 107-46-0, Hexamethyldisiloxane  
 111-30-8, Glutaraldehyde  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (biosensor using plasma-polymd. membrane)

IT 9059-11-4, Amine oxidase  
 RL: ARG (Analytical reagent use); DEV (Device component use); PEP  
 (Physical, engineering or chemical process); ANST (Analytical study); PROC  
 (Process); USES (Uses)  
 (biosensor using plasma-polymd. membrane)

RN 9059-11-4 HCPLUS

CN Oxidase, amine (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

IT 7440-06-4, Platinum, uses  
 RL: DEV (Device component use); USES (Uses)  
 (biosensor using plasma-polymd. membrane)

RN 7440-06-4 HCPLUS

CN Platinum (8CI, 9CI) (CA INDEX NAME)

Pt

L77 ANSWER 15 OF 33 HCPLUS COPYRIGHT 2003 ACS  
AN 2000:550147 HCPLUS  
DN 133:349278  
TI Amine oxidase-based flow biosensor for the assessment of fish freshness  
AU Frebort, Ivo; Skoupa, Lenka; Pec, Pavel  
CS Department of Biochemistry, Faculty of Science, Palacky University, Olomouc, 783 71, Czech Rep.  
SO Food Control (2000), 11(1), 13-18  
CODEN: FOOCEV; ISSN: 0956-7135  
PB Elsevier Science Ltd.  
DT Journal  
LA English  
CC 17-1 (Food and Feed Chemistry)  
AB Amine oxidases (EC 1.4.3.6) from grass pea (*Lathyrus sativus*) seedlings and fungus *Aspergillus niger* were immobilized to construct flow enzyme reactors for amine assay with spectrophotometric detection of enzymically produced hydrogen peroxide by a peroxidase/guaiacol system. While immobilized amine oxidase from *A. niger* showed poor storage stability, the *L. sativus* enzyme-based system was found useful for assay of putrefactive amines (putrescine and histamine) as markers of fish meat decompn. The optimized biosensor with av. lifetime 20 days showed a linear response to the amt. of histamine in the range 70-90 nmol with the assay limit of 4.4 nmol and putrescine in the range 0.9-70 nmol with the assay limit of 0.5 nmol.  
ST immobilized amine oxidase histamine analysis trout  
IT Food analysis  
Oncorhynchus mykiss  
(amine oxidase-based flow biosensor for the assessment of fish freshness)  
IT Enzymes, uses  
RL: ARG (Analytical reagent use); CAT (Catalyst use); ANST (Analytical study); USES (Uses)  
(immobilized; amine oxidase-based flow biosensor for the assessment of fish freshness)  
IT 51-45-6, Histamine, analysis 110-60-1, Putrescine  
RL: ANT (Analyte); ANST (Analytical study)  
(amine oxidase-based flow biosensor for the assessment of fish freshness)  
IT 9059-11-4, Amine oxidase  
RL: ARG (Analytical reagent use); CAT (Catalyst use); ANST (Analytical study); USES (Uses)  
(amine oxidase-based flow biosensor for the assessment of fish freshness)  
RE.CNT 19 THERE ARE 19 CITED REFERENCES AVAILABLE FOR THIS RECORD  
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IT 9059-11-4, Amine oxidase

RL: ARG (Analytical reagent use); CAT (Catalyst use); ANST (Analytical study); USES (Uses)  
 (amine oxidase-based flow biosensor for the assessment of fish freshness)

RN 9059-11-4 HCPLUS

CN Oxidase, amine (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

- L77 ANSWER 16 OF 33 · HCPLUS COPYRIGHT 2003 ACS  
 AN 2000:248465 HCPLUS  
 DN 133:116862  
 TI Amine oxidase based amperometric biosensors for histamine detection  
 AU Niculescu, Mihaela; Frebort, Ivo; Pec, Pavel;  
 Galuszka, Petr; Mattiasson, Bo; Csoregi, Elisabeth  
 CS Department of Biotechnology, Lund University, Lund, S-22100, Swed.  
 SO Electroanalysis (2000); 12(5), 369-375  
 CODEN: ELANEU; ISSN: 1040-0397  
 PB Wiley-VCH Verlag GmbH  
 DT Journal  
 LA English  
 CC 9-1 (Biochemical Methods)  
 AB This work reports on the development and optimization of amperometric biosensors based on the enzyme amine oxidase (AO) for the detection of histamine, a well-known biomarker for food freshness. Biosensor characteristics were evaluated in a flow injection (FI) anal. line, operated at +200 mV (vs. Ag/AgCl/0.1 M KCl). Two different biosensor designs were considered, one based on adsorbed AO on graphite electrodes, the detection being based on a direct electron transfer (DET) mechanism, whereas the second one based on an Osbipyridine modified redox polymer using a mediated electron transfer (MET) pathway. Both electrode designs were able to detect histamine in .mu.M range, however, the [osmium(4,4'-dimethylbipyridine)2Cl]<sup>+/-</sup> complexed with poly(1-vinylimidazole) (PVI13-dmeOs) based electrodes showed superior characteristics with regard to stability, selectivity and linear range. These electrodes were characterized by a detection limit of 2.2 .mu.M (calcd. as three times the signal-to-noise ratio), a sensitivity of 6.8 mA M<sup>-1</sup> cm<sup>-2</sup>, a linear range of 10-200 .mu.M, and an operational stability of 20% response loss during 8 h of continuous operation at a sample throughput of 30 injections h<sup>-1</sup>.  
 ST amine oxidase amperometric biosensor  
 histamine; enzyme electrode histamine detn  
 IT Enzyme electrodes  
 (amperometric, histamine-selective; amine oxidase-based amperometric graphite electrodes for histamine detection)  
 IT Chlorides, uses  
 RL: DEV (Device component use); USES (Uses)  
 (complexes with dimethylbipyridine, osmium and poly(1-

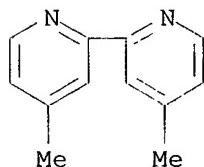
- vinylimidazole); amine oxidase-based amperometric graphite electrodes for histamine detection)
- IT 51-45-6, Histamine, analysis  
 RL: ANT (Analyte); ANST (Analytical study)  
 (amine oxidase-based amperometric graphite electrodes for histamine detection)
- IT 9059-11-4D, Amine oxidase, immobilized  
 RL: ARG (Analytical reagent use); DEV (Device component use); ANST (Analytical study); USES (Uses)  
 (amine oxidase-based amperometric graphite electrodes for histamine detection)
- IT 1134-35-6D, 4,4'-Dimethyl-2,2'-bipyridine, complexes with osmium, chloride and poly(1-vinylimidazole) 7440-04-2D, Osmium, complexes with dimethylbipyridine, chloride and poly(1-vinylimidazole), uses 25232-42-2D, Poly(1-vinylimidazole), complexes with osmium, dimethylbipyridine and chloride  
 RL: DEV (Device component use); USES (Uses)  
 (amine oxidase-based amperometric graphite electrodes for histamine detection)

RE.CNT 43 THERE ARE 43 CITED REFERENCES AVAILABLE FOR THIS RECORD

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 IT 9059-11-4D, Amine oxidase, immobilized  
 RL: ARG (Analytical reagent use); DEV (Device component use); ANST  
 (Analytical study); USES (Uses)  
 (amine oxidase-based amperometric graphite  
 electrodes for histamine detection)  
 RN 9059-11-4 HCAPLUS  
 CN Oxidase, amine (9CI) (CA INDEX NAME)
- \*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*  
 IT 1134-35-6D, 4,4'-Dimethyl-2,2'-bipyridine,  
 complexes with osmium, chloride and poly(1-  
 vinylimidazole) 7440-04-2D, Osmium, complexes  
 with dimethylbipyridine, chloride and poly(1-vinylimidazole),  
 uses 25232-42-2D, Poly(1-vinylimidazole), complexes  
 with osmium, dimethylbipyridine and chloride  
 RL: DEV (Device component use); USES (Uses)  
 (amine oxidase-based amperometric graphite  
 electrodes for histamine detection)  
 RN 1134-35-6 HCAPLUS  
 CN 2,2'-Bipyridine, 4,4'-dimethyl- (9CI) (CA INDEX NAME)



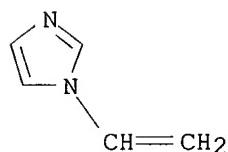
RN 7440-04-2 HCAPLUS  
 CN Osmium (8CI, 9CI) (CA INDEX NAME)

Os

RN 25232-42-2 HCAPLUS  
 CN 1H-Imidazole, 1-ethenyl-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 1072-63-5  
 CMF C5 H6 N2



L77 ANSWER 17 OF 33 HCAPLUS COPYRIGHT 2003 ACS  
 AN 2000:145190 HCAPLUS  
 DN 132:292859  
 TI Redox Hydrogel-Based Amperometric Bienzyme

AU      **Electrodes for Fish Freshness Monitoring**  
AU      Niculescu, Mihaela; Nistor, Catalin; Frebort, Ivo;  
AU      Pec, Pavel; Mattiasson, Bo; Csoeregi, Elisabeth  
CS      Department of Biotechnology, Lund University, Lund, S-22100, Swed.  
SO      Analytical Chemistry (2000); 72(7), 1591-1597  
SO      CODEN: ANCHAM; ISSN: 0003-2700  
PB      American Chemical Society  
DT      Journal  
LA      English  
CC      17-1 (Food and Feed Chemistry)  
AB      This work presents the design and optimization of amperometric biosensors for the detn. of biogenic amines (e.g., histamine, putrescine, cadaverine, tyramine, cystamine, agmatine, spermidine), commonly present in food products, and their application for monitoring of freshness in fish samples. The biosensors were used as the working electrodes of a three-electrode electrochem. cell of wall-jet type, operated at -50 mV vs. Ag/AgCl, in a flow injection system. Two different bienzyme electrode designs were considered, one based on the two enzymes [a newly isolated and purified amine oxidase (AO) and horseradish peroxidase (HRP)] simply adsorbed onto graphite electrodes, and one when they were cross-linked to an Os-based redox polymer. The redox hydrogel-based biosensors showed better biosensors characteristics, i.e., sensitivity of 0.194 A M-1 cm-2 for putrescine and 0.073 A M-1 cm-2 for histamine, and detection limits (calcd. as three times the signal-to-noise ratio) of 0.17 .mu.M for putrescine and 0.33 .mu.M for histamine. The optimized redox hydrogel-based biosensors were evaluated in terms of stability and selectivity, and were used for the detn. of total amine content in fish samples kept for 10 days in different conditions.  
ST      amperometric enzyme electrode amine detn fish; fish freshness monitoring amperometric biosensor  
IT      Food analysis  
      (amperometric bienzyme electrodes for detg. biogenic amines in)  
IT      Enzyme electrodes  
      (amperometric; redox hydrogel-based amperometric bienzyme electrodes for detg. biogenic amines in fish freshness monitoring)  
IT      Amines, analysis  
      RL: ANT (Analyte); ANST (Analytical study)  
      (biogenic; redox hydrogel-based amperometric bienzyme electrodes for detg. biogenic amines in fish freshness monitoring)  
IT      Fish  
      Hydrogels  
      (redox hydrogel-based amperometric bienzyme electrodes for detg. biogenic amines in fish freshness monitoring)  
IT      51-45-6, Histamine, analysis 51-67-2, Tyramine 51-85-4, Cystamine 107-15-3, Ethylenediamine, analysis 110-60-1, Putrescine 124-20-9, Spermidine  
      RL: ANT (Analyte); ANST (Analytical study)  
      (redox hydrogel-based amperometric bienzyme electrodes for detg. biogenic amines in fish freshness monitoring)  
IT      9003-99-0, Peroxidase 9059-11-4, Amine oxidase  
      RL: ARG (Analytical reagent use); DEV (Device component use); ANST (Analytical study); USES (Uses)  
      (redox hydrogel-based amperometric bienzyme electrodes for detg. biogenic amines in fish freshness

monitoring)

IT 9033-82-3D, complexes with osmium compd. 115304-16-0D,  
 complexes with vinylimidazole polymer  
 RL: DEV (Device component use); USES (Uses)  
 (redox hydrogel-based amperometric bienzyme  
 electrodes for detg. biogenic amines in fish freshness  
 monitoring)

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- (40) Vijayakumar, A; Anal Chim Acta 1996, V327, P223 HCPLUS
- (41) Volpe, G; Talanta 1996, V43, P283 HCPLUS
- (42) Xu, C; Talanta 1997, V44, P1625 HCPLUS
- (43) Yang, X; Electroanalysis 1995, V7, P105 HCPLUS
- (44) Yano, Y; Anal Chim Acta 1996, V320, P269 HCPLUS
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IT 9003-99-0, Peroxidase 9059-11-4, Amine

oxidase

RL: ARG (Analytical reagent use); DEV (Device component use); ANST (Analytical study); USES (Uses)  
 (redox hydrogel-based amperometric bienzyme  
 electrodes for detg. biogenic amines in fish freshness  
 monitoring)

RN 9003-99-0 HCPLUS

CN Peroxidase (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

RN 9059-11-4 HCAPLUS  
CN Oxidase, amine (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L77 ANSWER 18 OF 33 HCAPLUS COPYRIGHT 2003 ACS  
AN 2000:17220 HCAPLUS

DN 132:219076

TI Sensitive amperometric biosensor for the determination of biogenic and synthetic amines using pea seedlings amine oxidase: a novel approach for enzyme immobilisation  
AU Wimmerova, M.; Macholan, L.  
CS Department of Biochemistry, Faculty of Science, Masaryk University, Brno, 611 37, Czech Rep.  
SO Biosensors & Bioelectronics (1999), 14(8-9), 695-702  
CODEN: BBIOE4; ISSN: 0956-5663

PB Elsevier Science S.A.

DT Journal

LA English

CC 9-7 (Biochemical Methods)

AB We prep'd. a new inorg. sorbent based on modified triazine (2-[4,6-bis (aminoethylamine)-1,3,5-triazine]-Silasorb; BAT-Silasorb) which binds pea seedlings/amine oxidase (PSAO) very tightly without loss of its catalytic activity. This unique feature as well as the wide substrate specificity of PSAO was successfully utilized in the construction of an amperometric biosensor based on a carbon paste electrode for the fast and sensitive detection of various amines at a formal potential 0 mV vs. Ag/AgCl ref. electrode. The reaction layer of the biosensor is created by the direct immobilization of PSAO at the electrode surface via affinity carrier BAT-Silasorb. Used arrangement facilitates a simple restoration of the inactive biosensor. An amperometric signal results from horseradish peroxidase catalyzed redn. of H<sub>2</sub>O<sub>2</sub>, a secondary product of the oxidative deamination of amines, catalyzed by PSAO. The sensor was used for the basic characterization of 55 biogenic and synthetic amines, from numerous mono-, di- and polyamines to various hydroxy-, thio-, benzyl- and arom. derivs. in order to establish its suitability as a postcolumn detector. Its high sensitivity to putrescine 20.0.+-0.64 mA l-1 per mol (636.9.+-2.03 mA l-1 per mol per cm<sup>2</sup>), a limit of detection of 10 nmol l-1 (detd. with respect to a signal-to-noise ratio 3:1), a linear range of current response to 0.01-100 .mu.mol l-1 concn. of substrate and good reproducibility all indicate that the sensor could be applied to future industrial and clin. analyses.

ST biosensor amine detn; electrode enzyme amino oxidase amine detn; immobilization amino oxidase electrode

IT Amines, analysis

Amino acids, analysis

RL: ANT (Analyte); ANST (Analytical study)  
(amperometric carbon paste electrode for  
detn. of biogenic and synthetic amines using immobilized amine oxidase)

IT Enzyme electrodes

(amperometric, immobilized amine oxidase;  
amperometric carbon paste electrode for  
detn. of biogenic and synthetic amines using immobilized amine oxidase)

IT Monoamines

RL: ANT (Analyte); ANST (Analytical study)

- (biogenic, thiomonoamines and hydroxymonoamines; amperometric carbon paste electrode for detn. of biogenic and synthetic amines using pea seedlings amine oxidase)
- IT Paste electrodes  
 (carbon; amperometric carbon paste electrode for detn. of biogenic and synthetic amines using immobilized amine oxidase)
- IT Silica gel, analysis  
 RL: ARU (Analytical role, unclassified); DEV (Device component use); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses)  
 (conjugate with triazine deriv.; amperometric carbon paste electrode for detn. of biogenic and synthetic amines using pea seedlings amine oxidase)
- IT Amines, analysis  
 RL: ANT (Analyte); ANST (Analytical study)  
 (diamines; amperometric carbon paste electrode for detn. of biogenic and synthetic amines using pea seedlings amine oxidase)
- IT Immobilization, biochemical  
 (enzyme; amperometric carbon paste electrode for detn. of biogenic and synthetic amines using pea seedlings amine oxidase)
- IT Amines, analysis  
 RL: ANT (Analyte); ANST (Analytical study)  
 (polyamines, nonpolymeric; amperometric carbon paste electrode for detn. of biogenic and synthetic amines using pea seedlings amine oxidase)
- IT 51-45-6, Histamine, analysis 51-67-2, Tyramine 64-04-0,  
 Benzenethanamine 100-46-9, Benzylamine, analysis 120-20-7,  
 Homoveratrylamine  
 RL: ANT (Analyte); ANST (Analytical study)  
 (amperometric carbon paste electrode for detn. of biogenic and synthetic amines using immobilized amine oxidase)
- IT 7722-84-1, Hydrogen peroxide, analysis  
 RL: ANT (Analyte); ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)  
 (amperometric carbon paste electrode for detn. of biogenic and synthetic amines using immobilized amine oxidase)
- IT 51-85-4, Cystamine 56-87-1, L-Lysine, analysis 60-23-1, Cysteamine 70-47-3, L-Asparagine, analysis 70-54-2, Lysine 71-44-3, Spermine 74-89-5, Methylamine, analysis 75-31-0, Isopropylamine, analysis 78-81-9, Isobutylamine 96-20-8, 2-Amino-1-butanol 104-84-7, p-Methylbenzylamine 107-10-8, 1-Aminopropane, analysis 107-85-7, Isoamylamine 110-58-7, 1-Aminopentane 110-60-1, 1,4-Diaminobutane 111-26-2, 1-Aminohexane 111-68-2, 1-Aminoheptane 111-86-4, 1-Aminooctane 124-09-4, 1,6-Diaminohexane, analysis 124-20-9, Spermidine 156-87-6, 3-Amino-1-propanol 459-73-4, Glycine ethyl ester 462-94-2, Cadaverine 539-48-0, p-Xylylenediamine 539-59-3, 2-Hydroxyputrescine 540-27-2 590-88-5, 1,3-Diaminobutane 616-29-5, 2-Hydroxy-1,3-diaminopropane 1477-55-0, m-Xylylenediamine 1904-78-5, o-Nitrobenzylamine 4048-33-3, 6-Amino-1-hexanol 4117-33-3, L-Lysine ethyl ester 4403-69-4, o-Aminobenzylamine 4403-70-7 4403-71-8, p-Aminobenzylamine 7409-18-9 7409-30-5, p-Nitrobenzylamine 17061-62-0 17300-02-6, o-Xylylenediamine 19293-58-4, p-Dimethylaminobenzylamine 24177-21-7 32798-38-2 38595-00-5, 3-Hydroxycadaverine 40930-37-8 128505-66-8  
 RL: ANT (Analyte); PRP (Properties); ANST (Analytical study)  
 (amperometric carbon paste electrode for detn. of biogenic and synthetic amines using immobilized amine oxidase)

oxidase)

IT 107-15-3, 1,2-Diaminoethane, analysis  
 RL: ANT (Analyte); PRP (Properties); RCT (Reactant); ANST (Analytical study); RACT (Reactant or reagent)  
 (amperometric carbon paste electrode for  
 detn. of biogenic and synthetic amines using immobilized amine  
 oxidase)

IT 9003-99-0, Peroxidase  
 RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)  
 (amperometric carbon paste electrode for  
 detn. of biogenic and synthetic amines using immobilized amine  
 oxidase)

IT 9059-11-4, Amine oxidase  
 RL: ARG (Analytical reagent use); DEV (Device component use); PEP  
 (Physical, engineering or chemical process); ANST (Analytical study); PROC  
 (Process); USES (Uses)  
 (amperometric carbon paste electrode for  
 detn. of biogenic and synthetic amines using immobilized amine  
 oxidase)

IT 103658-99-7DP, conjugate with Silasorb  
 RL: ARU (Analytical role, unclassified); DEV (Device component use); SPN  
 (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES  
 (Uses)  
 (amperometric carbon paste electrode for  
 detn. of biogenic and synthetic amines using pea seedlings  
 amine oxidase)

IT 108-77-0, Cyanuric chloride 162164-08-1, Silasorb-Amine  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (amperometric carbon paste electrode for  
 detn. of biogenic and synthetic amines using pea seedlings  
 amine oxidase)

RE.CNT 22 THERE ARE 22 CITED REFERENCES AVAILABLE FOR THIS RECORD

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IT 9003-99-0, Peroxidase

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)  
 (amperometric carbon paste electrode for  
 detn. of biogenic and synthetic amines using immobilized amine  
 oxidase)

RN 9003-99-0 HCPLUS

CN Peroxidase (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

IT 9059-11-4, Amine oxidase

RL: ARG (Analytical reagent use); DEV (Device component use); PEP  
 (Physical, engineering or chemical process); ANST (Analytical study); PROC  
 (Process); USES (Uses)  
 (amperometric carbon paste electrode for  
 detn. of biogenic and synthetic amines using immobilized amine  
 oxidase)

RN 9059-11-4 HCPLUS

CN Oxidase, amine (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L77 ANSWER 19 OF 33 HCPLUS COPYRIGHT 2003 ACS

AN 1998:581464 HCPLUS

DN 129:270721

TI Design and development of an amperometric biosensor for  
 acetylcholine determination in brain microdialyzates

AU Larsson, N.; Ruzgas, T.; Gorton, L.; Kokai, M.; Kissinger, P.;  
 Csoregi, E.

CS Dep. Anal. Chem., Lund Univ., Lund, SE-221 00, Swed.

SO Electrochimica Acta (1998), 43(23), 3541-3554

CODEN: ELCAAV; ISSN: 0013-4686

PB Elsevier Science Ltd.

DT Journal

LA English

CC 2-1 (Mammalian Hormones)

AB An amperometric three-enzyme based biosensor for detn. of  
 acetylcholine has been developed with possible use for monitoring of brain  
 microdialyzates by co-immobilizing acetylcholinesterase (AchE), choline  
 oxidase (ChOx) and horseradish peroxidase

(HRP) in an Os-based redox polymer on solid  
 graphite electrodes. The redox hydrogel was  
 formed by crosslinking the appropriate enzymes and the/Os/  
 polymer (PVI13-dmeOs) working as a non-diffusing mediator between  
 the electrode and HRP. The sensor was used in a flow injection system at  
 an applied potential of -50 mV vs. Ag/AgCl. A detection limit  
 of 0.3 .mu.M (twice the S/N ratio) for acetylcholine was obtained, thus  
 representing a sensitive detection system. By adapting the electrode into  
 a microsystem, the release of acetylcholine in real samples (rat brain  
 dialyzates) could be shown. Electrode design, optimization steps and  
 characteristics for the optimized electrode configuration are presented.

ST amperometric biosensor acetylcholine brain microdialyzate

IT Biosensors

(amperometric; design and development of amperometric  
 biosensor for acetylcholine detn. in brain microdialyzates)

IT Brain

Electrodes

Flow injection systems

(design and development of amperometric biosensor for  
 acetylcholine detn. in brain microdialyzates)

IT Hydrogels

(redox; design and development of amperometric  
 biosensor for acetylcholine detn. in brain microdialyzates)

IT 51-84-3, Acetylcholine, analysis

RL: ANT (Analyte); ANST (Analytical study)

(design and development of amperometric biosensor for  
 acetylcholine detn. in brain microdialyzates)

IT 9000-81-1, Acetylcholinesterase 9028-67-5, Choline oxidase

RL: ARG (Analytical reagent use); DEV (Device component use); ANST  
 (Analytical study); USES (Uses)

(design and development of amperometric biosensor for  
 acetylcholine detn. in brain microdialyzates)

IT 7440-04-2, Osmium, uses

RL: DEV (Device component use); USES (Uses)  
 (design and development of amperometric biosensor for  
 acetylcholine detn. in brain microdialyzates)

IT 9003-99-0, Peroxidase

RL: ARG (Analytical reagent use); DEV (Device component use); ANST  
 (Analytical study); USES (Uses)  
 (horseradish; design and development of amperometric  
 biosensor for acetylcholine detn. in brain microdialyzates)

RE.CNT 44 THERE ARE 44 CITED REFERENCES AVAILABLE FOR THIS RECORD

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IT 7440-04-2, Osmium, uses

RL: DEV (Device component use); USES (Uses)  
 (design and development of amperometric biosensor for  
 acetylcholine detn. in brain microdialyzates)

RN 7440-04-2 HCPLUS

CN Osmium (8CI, 9CI) (CA INDEX NAME)

Os

IT 9003-99-0, Peroxidase

RL: ARG (Analytical reagent use); DEV (Device component use); ANST  
 (Analytical study); USES (Uses)  
 (horseradish; design and development of amperometric  
 biosensor for acetylcholine detn. in brain microdialyzates)

RN 9003-99-0 HCAPLUS

CN Peroxidase (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L77 ANSWER 20 OF 33 HCAPLUS COPYRIGHT 2003 ACS

AN 1998:374690 HCAPLUS

DN 129:158698

TI Enzyme biosensors based on electron transfer between electrode  
 and immobilized peroxidasesAU Gorton, Lo; Csoregi, Elisabeth; Ruzgas, Tautgirdas; Gazaryan,  
 Irina; Marko-Varga, GyorgyCS Department of Analytical Chemistry, Chemical Center, Lund University,  
 Lund, Swed.SO Methods in Biotechnology (1998), 6(Enzyme and Microbial Biosensors),  
 93-120

CODEN: MEBIFQ

PB Humana Press Inc.

DT Journal

LA English

CC 9-7 (Biochemical Methods)

AB The principle and construction of the title electrode are discussed.

ST enzyme biosensor electrode immobilized peroxidase

IT Electron transfer

Enzyme electrodes

(enzyme biosensors based on electron transfer between  
 electrode and immobilized peroxidases)

IT 9003-99-0, Peroxidase

RL: ARG (Analytical reagent use); DEV (Device component use); ANST  
 (Analytical study); USES (Uses)

(enzyme biosensors based on electron transfer between  
 electrode and immobilized peroxidases)

IT 9003-99-0, Peroxidase

RL: ARG (Analytical reagent use); DEV (Device component use); ANST  
 (Analytical study); USES (Uses)

(enzyme biosensors based on electron transfer between  
 electrode and immobilized peroxidases)

RN 9003-99-0 HCAPLUS

CN Peroxidase (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L77 ANSWER 21 OF 33 HCAPLUS COPYRIGHT 2003 ACS

AN 1998:86392 HCAPLUS

DN 128:241488

TI The main factors of monoamine biosensor selectivity increasing

AU Yagodina, Olga V.; Nikolskaya, Elena B.

CS Sechenov Institute of Evolutionary Physiology and Biochemistry, Russian  
 Academy of Sciences, St. Petersburg, 194223, Russia

SO Sensors and Actuators, B: Chemical (1997), B44(1-3), 566-570

CODEN: SABCEB; ISSN: 0925-4005

PB Elsevier Science S.A.

DT Journal

LA English

CC 9-16 (Biochemical Methods)  
 Section cross-reference(s): 7

AB The influence of the main factors of **biosensor** selectivity on monoamine detn. have been studied. In the compn. of new **biosensors, amine oxidases** (AO) from different sources were used: mitochondrial AO from pig and rat liver and AO from Methanoscoccus barkeri strain 27. Enzyme preps. of different degrees of purifn. and immobilized in different ways have been studied. Potentiometric **electrodes**, gas-sensing **electrodes**, and colorimetric **sensors** were used as the anal. detectors in the designed **biosensors**. New methods for the individual detn. of monoamines and for the detection of their sum in the sample have been worked out.

ST monoamine **biosensor** selectivity **amine oxidase electrode**

IT Enzyme **electrodes**  
 (gas-sensing; main factors increasing monoamine **biosensor** selectivity)

IT **Biosensors**  
 (main factors increasing monoamine **biosensor** selectivity)

IT Monoamines  
 RL: ANT (Analyte); BPR (Biological process); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study); PROC (Process)  
 (main factors increasing monoamine **biosensor** selectivity)

IT Enzyme **electrodes**  
 (potentiometric; main factors increasing monoamine **biosensor** selectivity)

IT Immobilization, biochemical  
 (protein, **amine oxidase**; main factors increasing monoamine **biosensor** selectivity)

IT Gelatins, uses  
 RL: NUU (Other use, unclassified); USES (Uses)  
 (use in **amine oxidase** immobilization; main factors increasing monoamine **biosensor** selectivity)

IT 50-67-9, Serotonin, analysis 51-67-2, Tyramine 100-46-9, Benzylamine, analysis  
 RL: ANT (Analyte); BPR (Biological process); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study); PROC (Process)  
 (main factors increasing monoamine **biosensor** selectivity)

IT 9059-11-4D, Amine **oxidase**, immobilized  
 RL: ARG (Analytical reagent use); BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study); USES (Uses)  
 (main factors increasing monoamine **biosensor** selectivity)

IT 9002-18-0D, Agar, **amine oxidase** conjugate  
 RL: NUU (Other use, unclassified); USES (Uses)  
 (use in **amine oxidase** immobilization; main factors increasing monoamine **biosensor** selectivity)

IT 9059-11-4D, Amine **oxidase**, immobilized  
 RL: ARG (Analytical reagent use); BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study); USES (Uses)  
 (main factors increasing monoamine **biosensor** selectivity)

RN 9059-11-4 HCPLUS

CN Oxidase, amine (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

TI Preliminary investigations of an amperometric oligosaccharide dehydrogenase-based electrode for the detection of glucose and some other low molecular weight saccharides

AU Ruzgas, T.; Csoregi, E.; Katakis, I.; Kenausis, G.; Gorton, L.

CS Enzyme Chem. Lab., Inst. Biochem., Vilnius, 26000, Lithuania

SO Journal of Molecular Recognition (1996), 9(5/6), 480-484

CODEN: JMOR4; ISSN: 0952-3499

PB Wiley

DT Journal

LA English

CC 9-7 (Biochemical Methods)

AB Biosensors for the detn. of sugars were constructed using oligosaccharide dehydrogenase (ODH) and sol. phenazine methosulfate (PMS) or an osmium-based three-dimensional redox hydrogel. In the latter case the enzyme and poly(1-vinyylimidazole) complexed with osmium (4,4'-dimethylbpy)2Cl were cross-linked with poly(ethylene glycol) diglycidyl ether.

Both electrode configurations showed similar sensitivities for glucose in the range between 8 and 21 .mu.AmM-1 cm-2. The responses for 10 mono and oligosaccharides were studied. There was no response for fructose. In the concn. range 0.1-2.0 mM the relative sensitivities were detd. for arabinose (96%), xylose (3%), mannose (50%), galactose (11%), glucose (100%), maltose (24%), lactose (12%), cellobiose (34%) and maltotriose (10%).

ST glucose saccharide detn oligosaccharide dehydrogenase electrode

IT Electrodes

(preliminary investigations of an amperometric oligosaccharide dehydrogenase-based electrode for the detection of glucose and some other low mol. wt. saccharides)

IT Carbohydrates, analysis

RL: ANT (Analyte); ANST (Analytical study)

(preliminary investigations of an amperometric oligosaccharide dehydrogenase-based electrode for the detection of glucose and some other low mol. wt. saccharides)

IT 50-99-7, Glucose, analysis 58-86-6, Xylose, analysis 59-23-4, Galactose, analysis 63-42-3, Lactose 69-79-4, Maltose 147-81-9, Arabinose 528-50-7, Cellobiose 1109-28-0, Maltotriose 3458-28-4, Mannose

RL: ANT (Analyte); ANST (Analytical study)

(preliminary investigations of an amperometric oligosaccharide dehydrogenase-based electrode for the detection of glucose and some other low mol. wt. saccharides)

IT 299-11-6, Phenazine methosulfate 122191-33-7, Oligosaccharide dehydrogenase

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)

(preliminary investigations of an amperometric oligosaccharide dehydrogenase-based electrode for the detection of glucose and some other low mol. wt. saccharides)

L77 ANSWER 23 OF 33 HCAPLUS COPYRIGHT 2003 ACS

AN 1997:49634 HCAPLUS

TI Amperometric biosensor with immobilized pea seedlings/  
amine oxidase

AU Wimmerova, Michaela; Macholan, Lumir

CS Dep. Biochem., Masaryk Univ., Brno, 611 37, Czech Rep.

SO Chem. Listy (1996), 90(9), 725

CODEN: CHLSAC; ISSN: 0009-2770

PB Ceska Spolecnost Chemicka

DT Journal

LA Czech

AB Unavailable

L77 ANSWER 24 OF 33 HCPLUS COPYRIGHT 2003 ACS  
 AN 1997:9160 HCPLUS  
 DN 126:28836  
 TI Ezzyme electrode  
 IN Karube, Masao; Nagata, Ryohei  
 PA Karube Masao, Japan; Dainippon Printing Co Ltd  
 SO Jpn. Kokai Tokkyo Koho, 16 pp.  
 CODEN: JKXXAF  
 DT Patent  
 LA Japanese  
 IC ICM G01N027-327  
 CC 9-7 (Biochemical Methods)  
 Section cross-reference(s): 72

## FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 08271472	A2	19961018	JP 1995-93171	19950328
PRAI	JP 1995-93171		19950328		

AB Disclosed is compn. comprising conductive enzyme, metal complex, nicotinamide deriv., flavin deriv. quinone or quinone deriv., hydrophilic and/or hydrophobic **polymer** for prepn. of enzyme **sensor** for anal. The enzyme **electrode** is useful for rapid detection of analyte in bio-sample.

ST enzyme **electrode** metal complex flavin nicotinamide

IT Metals, uses

RL: DEV (Device component use); USES (Uses)  
 (complex; enzyme **electrode** comprises conductive enzyme, metal complex, and nicotinamide deriv)

IT Enzymes, uses

RL: DEV (Device component use); USES (Uses)  
 (conductive; enzyme **electrode** comprises conductive enzyme, metal complex, and nicotinamide deriv)

IT Flavins

RL: DEV (Device component use); USES (Uses)  
 (deriv.; enzyme **electrode** comprises conductive enzyme, metal complex, and nicotinamide deriv)

IT Enzyme **electrodes**

(enzyme **electrode** comprises conductive enzyme, metal complex, and nicotinamide deriv)

IT Polyvinyl butyrals

Sandwich compounds

RL: DEV (Device component use); USES (Uses)  
 (enzyme **electrode** comprises conductive enzyme, metal complex, and nicotinamide deriv)

IT Biosensors

(enzymic; enzyme **electrode** comprises conductive enzyme, metal complex, and nicotinamide deriv)

IT Polymers, uses

RL: DEV (Device component use); USES (Uses)  
 (hydrophobic and/or hydrophilic; enzyme **electrode** comprises conductive enzyme, metal complex, and nicotinamide deriv)

IT 102-54-5, Ferrocene

RL: DEV (Device component use); USES (Uses)  
 (deriv.; enzyme **electrode** comprises conductive enzyme, metal complex, and nicotinamide deriv)

IT 98-92-0D, 3-Pyridinecarboxamide, deriv. 106-51-4D, 2,5-Cyclohexadiene-1,4-dione, deriv. 1071-93-8 4080-95-9 9000-88-8 9000-89-9  
 9001-37-0 9001-96-1 9003-39-8 9028-67-5 9028-76-6 9028-79-9  
 9035-73-8, Oxidase 9059-11-4 13043-98-6 14323-06-9

20247-84-1 23570-43-6 33037-04-6 34796-67-3 64616-77-9

RL: DEV (Device component use); USES (Uses)  
 (enzyme **electrode** comprises conductive enzyme, metal complex, and nicotinamide deriv)

IT 9059-11-4  
 RL: DEV (Device component use); USES (Uses)  
 (enzyme electrode comprises conductive enzyme, metal complex,  
 and nicotinamide deriv)  
 RN 9059-11-4 HCAPLUS  
 CN Oxidase, amine (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L77 ANSWER 25 OF 33 HCAPLUS COPYRIGHT 2003 ACS

AN 1996:252397 HCAPLUS

DN 124:283713

TI Composition for enzyme electrode

IN Watanabe, Masayoshi

PA Dainippon Printing Co Ltd, Japan

SO Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G01N027-327

ICS C08F220-28; C08F230-04

CC 9-7 (Biochemical Methods)

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 08029372	A2	19960202	JP 1994-232508	19940902
PRAI	JP 1994-117392		19940509		

AB Compn. contg. enzyme, **polymeric** mediator, conductive component, binder, etc. is disclosed for prep. enzyme **biosensor** with wide detection spectrum, high sensitivity, and long-life. The mediator is a **homopolymer** or **copolymer** of redox-active monomer, e.g. derivs. of ferrocene, nicotinamide, flavin, quinone, etc. The enzyme is an **oxidase** or dehydrogenase; and the conductive component is a metal and/or **carbon** microparticle. In example, vinylferrocene-methoxynanoethylene oxide methacrylate **copolymer** was prep'd. as mediator, mixed with glucose **oxidase**, and coated on **electrode** for glucose detn. Similarly, enzyme **electrode** contg. vinylferrocene-dodecyl methacrylate as mediator was also prep'd. for the same purpose.

ST enzyme electrode **polymer copolymer** mediator

IT Enzyme electrodes

(compn. contg. enzyme and **polymeric** mediator and conductive component and binder, for enzyme **electrode**)

IT Enzymes

RL: DEV (Device component use); USES (Uses)

(compn. contg. enzyme and **polymeric** mediator and conductive component and binder, for enzyme **electrode**)

IT Metals

RL: DEV (Device component use); USES (Uses)

(conductive; compn. contg. enzyme and **polymeric** mediator and conductive component and binder, for enzyme **electrode**)

IT Flavins

RL: DEV (Device component use); USES (Uses)

(derivs.; **polymer** or **copolymer**; compn. contg.

enzyme and **polymeric** mediator and conductive component and binder, for enzyme **electrode**)

IT Biosensors

(enzyme; compn. contg. enzyme and **polymeric** mediator and conductive component and binder, for enzyme **electrode**)

IT Polymers

RL: DEV (Device component use); USES (Uses)

(mediator; compn. contg. enzyme and **polymeric** mediator and conductive component and binder, for enzyme **electrode**)

## IT Polymers

RL: DEV (Device component use); USES (Uses)  
 (co-, mediator; compn. contg. enzyme and **polymeric** mediator  
 and conductive component and binder, for enzyme electrode)

IT 50-99-7, D-Glucose  
 RL: ANT (Analyte); ANST (Analytical study)  
 (compn. contg. enzyme and **polymeric** mediator and conductive  
 component and binder, for enzyme electrode)

IT 75-01-4D, derivs.; polymers 75-35-4D, derivs.;  
 polymers 78-79-5D, derivs.; polymers 79-10-7D,  
 2-Propenoic acid, derivs., polymers with ferrocenes 79-41-4D,  
 derivs., polymers with ferrocenes 98-83-9D, derivs.;  
 polymers 98-92-0D, 3-Pyridinecarboxamide, derivs.;  
 polymers 100-42-5D, derivs.; polymers 102-54-5D,  
 Ferrocene, derivs.; polymers 106-51-4D, 2,5-Cyclohexadiene-1,4-  
 dione, derivs.; polymers 106-99-0D, 1,3-Butadiene, derivs.;  
 polymers 108-05-4D, Acetic acid ethenyl ester, derivs.;  
 polymers 115-11-7D, derivs.; polymers 9000-88-8  
 9000-89-9 9001-37-0 9001-96-1 9028-14-2 9028-21-1 9028-53-9  
 9028-67-5 9028-76-6 9028-79-9 9028-86-8 9031-72-5 9035-73-8,  
 Oxidase 9035-82-9, Dehydrogenase 9059-11-4  
 67775-34-2 135622-84-3 166274-80-2 175735-60-1

RL: DEV (Device component use); USES (Uses)  
 (compn. contg. enzyme and **polymeric** mediator and conductive  
 component and binder, for enzyme electrode)

IT 74-85-1D, Ethene, derivs.; polymers 107-13-1D,  
 2-Propenenitrile, derivs.; polymers 9001-46-1

RL: DEV (Device component use); USES (Uses)  
 (compn. contg. enzyme and **polymeric** mediator and conductive  
 component and binder, for enzyme electrode)

## IT 7440-44-0, Carbon

RL: DEV (Device component use); USES (Uses)  
 (microparticles; compn. contg. enzyme and **polymeric** mediator  
 and conductive component and binder, for enzyme electrode)

## IT 9059-11-4

RL: DEV (Device component use); USES (Uses)  
 (compn. contg. enzyme and **polymeric** mediator and conductive  
 component and binder, for enzyme electrode)

RN 9059-11-4 HCPLUS

CN Oxidase, amine (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

## IT 7440-44-0, Carbon

RL: DEV (Device component use); USES (Uses)  
 (microparticles; compn. contg. enzyme and **polymeric** mediator  
 and conductive component and binder, for enzyme electrode)

RN 7440-44-0 HCPLUS

CN Carbon (7CI, 8CI, 9CI) (CA INDEX NAME)

C

L77 ANSWER 26 OF 33 HCPLUS COPYRIGHT 2003 ACS  
 AN 1995:777885 HCPLUS  
 DN 123:164079  
 TI Sensor electrode containing immobilized enzymes and  
 hydrophilic or hydrophilic resins  
 IN Karube, Masao; Nagata, Ryohei  
 PA Karube Masao, Japan; Dainippon Printing Co Ltd  
 SO Jpn. Kokai Tokkyo Koho, 15 pp.  
 CODEN: JKXXAF

DT Patent  
 LA Japanese  
 IC ICM G01N027-327  
 CC 7-7 (Enzymes)  
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 07151727	A2	19950616	JP 1994-257272	19940928
PRAI	JP 1993-264298		19930928		
AB	<b>A sensor electrode</b> was constructed which contained hydrophilic resin (e.g. polyvinylpyrrolidone) or hydrophobic resin (e.g. polyvinylbutyral), an <b>oxidase</b> (e.g. glucose <b>oxidase</b> ), and an enzyme mediator (e.g. ferrocene, nicotine amine, and quinone). Glucose concn. was detd. by the <b>sensor electrode</b> contg. glucose <b>oxidase</b> . Examples of other <b>oxidases</b> are galactose <b>oxidase</b> , pyruvate <b>oxidase</b> , D- and L-amino acid <b>oxidase</b> , <b>amine oxidase</b> , cholesterol <b>oxidase</b> , and choline <b>oxidase</b> .				
ST	<b>sensor electrode</b> immobilization resin enzyme				
IT	Flavins RL: NUU (Other use, unclassified); USES (Uses) (enzyme mediator; <b>sensor electrode</b> contg. immobilized enzymes and hydrophilic or hydrophilic resins)				
IT	Resins RL: NUU (Other use, unclassified); USES (Uses) (hydrophilic and hydrophobic; <b>sensor electrode</b> contg. immobilized enzymes and hydrophilic or hydrophilic resins)				
IT	Enzymes RL: NUU (Other use, unclassified); USES (Uses) ( <b>sensor electrode</b> contg. immobilized enzymes and hydrophilic or hydrophilic resins)				
IT	Vinyl acetal polymers RL: NUU (Other use, unclassified); USES (Uses) (butyrals, <b>sensor electrode</b> contg. immobilized enzymes and hydrophilic or hydrophilic resins)				
IT	Sensors (electrochem., <b>sensor electrode</b> contg. immobilized enzymes and hydrophilic or hydrophilic resins)				
IT	50-99-7, Glucose, analysis RL: ANT (Analyte); ANST (Analytical study) (detn. of; <b>sensor electrode</b> contg. immobilized enzymes and hydrophilic or hydrophilic resins)				
IT	98-92-0, 3-Pyridinecarboxamide 102-54-5, Ferrocene 106-51-4, Quinone, uses 1271-42-7, Ferrocene carboxylic acid RL: NUU (Other use, unclassified); USES (Uses) (enzyme mediator; <b>sensor electrode</b> contg. immobilized enzymes and hydrophilic or hydrophilic resins)				
IT	9000-88-8, D-Amino acid <b>oxidase</b> 9000-89-9, L-Amino acid <b>oxidase</b> 9001-37-0, Glucose <b>oxidase</b> 9001-96-1, Pyruvate <b>oxidase</b> 9003-39-8, Polyvinylpyrrolidone 9028-67-5, Choline <b>oxidase</b> 9028-76-6, Cholesterol <b>oxidase</b> 9028-79-9, Galactose <b>oxidase</b> 9035-73-8, Oxidase 9059-11-4, <b>Amine oxidase</b> RL: NUU (Other use, unclassified); USES (Uses) ( <b>sensor electrode</b> contg. immobilized enzymes and hydrophilic or hydrophilic resins)				
IT	9059-11-4, <b>Amine oxidase</b> RL: NUU (Other use, unclassified); USES (Uses) ( <b>sensor electrode</b> contg. immobilized enzymes and hydrophilic or hydrophilic resins)				
RN	9059-11-4 HCPLUS				
CN	Oxidase, amine (9CI) (CA INDEX NAME)				

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L77 ANSWER 27 OF 33 HCPLUS COPYRIGHT 2003 ACS  
AN 1994:577990 HCPLUS  
DN 121:177990  
TI Oxygen-sensor-based simple assay of histamine in fish using purified amine oxidase  
AU Ohashi, Minoru; Nomura, Fumiko; Suzuki, Mieko; Otsuka, Megumi; Adachi, Osao; Arakawa, Nobuhiko  
CS Moritex Co., Tokyo, 150, Japan  
SO Journal of Food Science (1994), 59(3), 519-22  
CODEN: JFDSAZ; ISSN: 0022-1147  
DT Journal  
LA English  
CC 17-1 (Food and Feed Chemistry)  
AB Oxygen consumption was measured by an oxygen sensor after addn. of purified fungal amine oxidase to fish exts. The oxidn. of histamine to imidazole acetaldehyde proceeded stoichiometrically. Based on the equimolar relationship between histamine and oxygen consumption, histamine was detd. selectively by the oxygen sensor. Neither sample pretreatment removing interfering materials nor daily calibration by histamine std. was required. Histamine contents in scombroid fish were detd. rapidly with good accuracy. AOAC and oxygen sensor methods showed a very high correlation ( $r = 0.999$ ,  $n = 6$ ).  
ST histamine detn fish oxygen biosensor; amine oxidase histamine assay  
IT Biosensors  
(for oxygen, histamine detn. in fish with, amine oxidase in)  
IT Fish  
Mackerel  
Trachurus  
Tuna  
(histamine detn. in, by biosensor for oxygen, amine oxidase in)  
IT Tuna  
(canned, histamine detn. in, by biosensor for oxygen, amine oxidase in)  
IT Euthynnus affinis  
(frozen, histamine detn. in, by biosensor for oxygen, amine oxidase in)  
IT Canned foods  
Frozen foods  
(tuna, histamine detn. in, by biosensor for oxygen, amine oxidase in)  
IT 7782-44-7, Oxygen, miscellaneous  
RL: MSC (Miscellaneous)  
(biosensor for, histamine detn. in fish with, amine oxidase in)  
IT 51-45-6, Histamine, analysis  
RL: ANT (Analyte); ANST (Analytical study)  
(detn. of, in fish by biosensor for oxygen, amine oxidase in)  
IT 9059-11-4, Amine oxidase  
RL: ANST (Analytical study)  
(in histamine detn. in fish, by oxygen biosensor)  
IT 9059-11-4, Amine oxidase  
RL: ANST (Analytical study)  
(in histamine detn. in fish, by oxygen biosensor)  
RN 9059-11-4 HCPLUS  
CN Oxidase, amine (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L77 ANSWER 28 OF 33 HCAPLUS COPYRIGHT 2003 ACS  
 AN 1994:574216 HCAPLUS  
 DN 121:174216  
 TI Renewable miniature enzyme-based sensing devices  
 AU Gasparini, R.; Scarpa, M.; Vianello, F.; Mondovi, B.; Rigo, A.  
 CS Department of Biological Chemistry, University of Padova, Via Trieste 75,  
 Padova, 35100, Italy  
 SO Analytica Chimica Acta (1994), 294(3), 299-304  
 CODEN: ACACAM; ISSN: 0003-2670  
 DT Journal  
 LA English  
 CC 9-1 (Biochemical Methods)  
 AB A new approach to the prepn. of electrochem. biosensors, based  
 on a mixed Sepharose-carbon paste electrode,  
 is described. The bioelectrode is made from carbon  
 paste which is mixed, during prepn., with a mediator and with  
 Sepharose contg. an immobilized enzyme. The immobilized enzymes were  
 glucose oxidase, from Aspergillus niger, and amine  
 oxidase from bovine serum and from soybean seedlings.  
 The Sepharose environment, favorable to the enzyme, and the close  
 proximity of the enzyme redox-mediating and sensing sites, permits the  
 required amt. of enzyme to be decreased by two orders of magnitude and  
 allows rapid response to the substrate. Response times as short as 15 s  
 have been measured. The microelectrodes are easily fabricated,  
 and the modified carbon paste can be incorporated in  
 the various sensor configurations (micro, flow, etc.) relevant  
 to clin. anal.  
 ST renewable miniature enzyme based sensing device  
 IT Electrodes  
     (bio-, renewable miniature enzyme-based sensing devices)  
 IT Electrodes  
     (bio-, glucose-selective, renewable miniature enzyme-based sensing  
     devices)  
 IT 50-99-7, Glucose, analysis  
     RL: ANT (Analyte); ANST (Analytical study)  
     (renewable miniature enzyme-based sensing devices)  
 IT 9001-37-0D, Glucose oxidase, immobilized 9059-11-4D,  
     Amine oxidase, immobilized 58856-73-8, Ah sepharose  
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)  
     (renewable miniature enzyme-based sensing devices)  
 IT 9059-11-4D, Amine oxidase, immobilized  
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)  
     (renewable miniature enzyme-based sensing devices)  
 RN 9059-11-4 HCAPLUS  
 CN Oxidase, amine (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L77 ANSWER 29 OF 33 HCAPLUS COPYRIGHT 2003 ACS  
 AN 1994:100871 HCAPLUS  
 DN 120:100871  
 TI A reagentless amperometric biosensor for alcohol detection in  
 column liquid chromatography based on co-immobilized peroxidase  
 and alcohol oxidase in carbon paste  
 AU Johansson, K.; Joensson-Pettersson, G.; Gorton, L.; Marko-Varga, G.;  
 Csoregi, E.  
 CS Dep. Anal. Chem., Univ. Lund, Lund, S-221 00, Swed.  
 SO Journal of Biotechnology (1993), 31(3), 301-16  
 CODEN: JBITD4; ISSN: 0168-1656  
 DT Journal  
 LA English

CC 9-1 (Biochemical Methods)  
 Section cross-reference(s): 16, 72, 80

AB A reagentless **C paste** electrode chem. modified with covalently bound alc. **oxidase** and horseradish **peroxidase** was examd. as a selective sensor in flow injection and column liq. chromatog. A combination of carbodiimide, glutaraldehyde, and polyethylenimine was used for immobilizing the enzymes in the **paste**. The surface of the electrodes was protected by first forming a layer of **electropolymer**, o-phenylenediamine followed by deposition of a cation-exchange membrane (Eastman AQ 29D). The electrodes were used for detection of hydrogen peroxide, methanol, ethanol, propanol, isopropanol, and butanol. Preliminary investigations of the use of this sensor for bioprocess control are reported.

ST alc detection amperometric **biosensor** liq chromatog; **carbon paste** enzyme electrode alc detection

IT Alcohols, analysis  
 RL: ANT (Analyte); ANST (Analytical study)  
 (detection of, by liq. chromatog. with amperometric enzyme electrode)

IT Immobilization, biochemical  
 (of alc. **oxidase** and **peroxidase**, in **carbon paste** alc.-selective amperometric electrode)

IT Electrodes  
 (bio-, enzyme, alc.-selective, amperometric, **carbon paste**, in liq. chromatog. detector)

IT Chromatographs, column and liquid  
 (detectors, electrochem., amperometric alc.-selective enzyme electrode in)

IT 64-17-5, Ethanol, analysis 67-56-1, Methanol, analysis 67-63-0, Isopropanol, analysis 71-23-8, Propanol, analysis 71-36-3, Butanol, analysis 7722-84-1, Hydrogen peroxide, analysis  
 RL: ANT (Analyte); ANST (Analytical study)  
 (detection of, by liq. chromatog. with amperometric enzyme electrode)

IT 9003-99-0D, **Peroxidase**, immobilized 9073-63-6D,  
 Alcohol **oxidase**, immobilized  
 RL: ANST (Analytical study)  
 (in alc.-selective amperometric electrode for liq. chromatog. detection)

IT 7440-44-0, **Carbon**, uses  
 RL: USES (Uses)  
 (**paste**, electrode, with immobilized enzymes, for alc. detn.  
 by liq. chromatog.)

IT 9003-99-0D, **Peroxidase**, immobilized  
 RL: ANST (Analytical study)  
 (in alc.-selective amperometric electrode for liq. chromatog. detection)

RN 9003-99-0 HCAPLUS

CN Peroxidase (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

IT 7440-44-0, **Carbon**, uses  
 RL: USES (Uses)  
 (**paste**, electrode, with immobilized enzymes, for alc. detn.  
 by liq. chromatog.)

RN 7440-44-0 HCAPLUS

CN Carbon (7CI, 8CI, 9CI) (CA INDEX NAME)

C

DN 119:265773  
 TI Amperometric biosensors based on immobilized redox-enzymes in carbon paste electrodes  
 AU Gorton, L.; Dominguez, E.; Marko-Varga, G.; Persson, B.; Joensson-Pettersson, E.; Csoregi, E.; Johansson, K.; Narasaiah, D.; Ghobadi, S.  
 CS Dep. Anal. Chem., Univ. Lund, Lund, S-221 00, Swed.  
 SO Bioelectroanal., 2, Symp., 2nd (1993), Meeting Date 1992, 33-58.  
 Editor(s): Pungor, Erno. Publisher: Akad, Kiado, Budapest, Hung.  
 CODEN: 59LGAV  
 DT Conference  
 LA English  
 CC 9-7 (Biochemical Methods)  
 Section cross-reference(s): 7  
 AB A no. of redox enzymes have been immobilized in carbon paste electrodes operating around 0 mV vs. SCE. Examples are given of an alc. sensor based on alc. dehydrogenase, a fructose sensor based on fructose dehydrogenase, an L-lactate sensor based on co-immobilized L-lactate oxidase and a fungal peroxidase, and an L-glutamate sensor based on co-immobilized L-glutamate oxidase and horse radish peroxidase. The pos. effects on the sensor performances on the addn. of polyethyleneimine are demonstrated.  
 ST amperometric biosensor electrode redox enzyme  
 IT Immobilization, biochemical  
     (of redox enzymes on carbon paste electrodes in amperometric biosensor construction)  
 IT Electrodes  
     (bio-, enzyme, amperometric, paste, properties of, using redox enzymes)  
 IT Enzymes  
     RL: PROC (Process)  
         (redox, immobilization of, in amperometric biosensor electrode)  
 IT 9002-98-6  
     RL: ANST (Analytical study)  
         (redox enzyme properties in amperometric biosensor response to)

L77 ANSWER 31 OF 33 HCPLUS COPYRIGHT 2003 ACS  
 AN 1993:665610 HCPLUS  
 DN 119:265610  
 TI Miniature amperometric biosensors for detection of hydrogen peroxide and glucose based on peroxidase modified carbon fibers  
 AU Csoregi, Elisabeth; Gorton, Lo; Marko-Varga, Gyorgy  
 CS Dep. Analy. Chem., Univ. Lund, Lund, S-221 00, Swed.  
 SO Bioelectroanal., 2, Symp., 2nd (1993), Meeting Date 1992, 271-84.  
 Editor(s): Pungor, Erno. Publisher: Akad, Kiado, Budapest, Hung.  
 CODEN: 59LGAV  
 DT Conference  
 LA English  
 CC 9-1 (Biochemical Methods)  
 AB A reagentless miniature amperometric biosensor can be constructed for the detn. of glucose by co-immobilizing horse radish peroxidase with the H<sub>2</sub>O<sub>2</sub> producing glucose oxidase on C fibers. The detection is based on an apparent direct electron transfer between the electrode and the active center of the immobilized peroxidase. The detection can be made within the optimal potential range. The various optimization steps are described. A linear response range was obtained between 40-2500 .mu.M H<sub>2</sub>O<sub>2</sub>. Linear calibration curves for glucose were obtained between 20-160 .mu.M glucose. An av. conversion efficiency of glucose of 58% was calcd. as the ratio between the signal for glucose and for H<sub>2</sub>O<sub>2</sub> from the linear calibration

curves.

ST biosensor hydrogen peroxide glucose detection;  
peroxidase carbon fiber enzyme electrode

IT Carbon fibers, uses  
RL: USES (Uses)  
(electrode, peroxidase immobilization on, for hydrogen peroxide detection by biosensors)

IT Electrodes  
(bio-, enzyme, amperometric, with immobilized glucose oxidase and peroxidase, for hydrogen peroxide and glucose detection)

IT 7440-44-0  
RL: ANST (Analytical study)  
(carbon fibers, electrode, peroxidase immobilization on, for hydrogen peroxide detection by biosensors)

IT 50-99-7, Glucose, analysis 7722-84-1, Hydrogen peroxide, analysis  
RL: ANT (Analyte); ANST (Analytical study)  
(detection of, by peroxidase-contg. carbon fiber biosensors)

IT 9003-99-0, Peroxidase  
RL: PROC (Process)  
(immobilization of, on carbon fibers biosensor, for hydrogen peroxide detection)

IT 7440-44-0  
RL: ANST (Analytical study)  
(carbon fibers, electrode, peroxidase immobilization on, for hydrogen peroxide detection by biosensors)

RN 7440-44-0 HCAPLUS  
CN Carbon (7CI, 8CI, 9CI) (CA INDEX NAME)

C

IT 9003-99-0, Peroxidase  
RL: PROC (Process)  
(immobilization of, on carbon fibers biosensor, for hydrogen peroxide detection)

RN 9003-99-0 HCAPLUS  
CN Peroxidase (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L77 ANSWER 32 OF 33 HCAPLUS COPYRIGHT 2003 ACS  
AN 1992:644760 HCAPLUS  
DN 117:244760  
TI Amperometric biosensors based on an apparent direct electron transfer between electrodes and immobilized peroxidases  
AU Gorton, Lo; Joensson-Pettersson, Gunilla; Csoregi, Elisabeth; Johansson, Kristina; Dominguez, Elena; Marko-Varga, Gyorgy  
CS Dep. Anal. Chem., Univ. Lund, Lund, S-221 00, Swed.  
SO Analyst (Cambridge, United Kingdom) (1992), 117(8), 1235-41  
CODEN: ANALAO; ISSN: 0003-2654  
DT Journal  
LA English  
CC 80-2 (Organic Analytical Chemistry)  
Section cross-reference(s): 9  
AB An apparent direct electron transfer between various electrode materials and peroxidases immobilized on the surface of the electrode has been reported in the last few years. An electrocatalytic redn. of hydrogen peroxide starts at about +600 mV vs. a satd. calomel (ref.) electrode (SCE) at neutral pH. The efficiency of the electrocatalytic

current increases as the applied potential is made more neg. and starts to level off at about -200 mV vs. SCE. Amperometric biosensors for hydrogen peroxide can be constructed with these types of **peroxidase** modified electrodes. By co-immobilizing a hydrogen peroxide-producing **oxidase** with the **peroxidase**, amperometric biosensors can be made that respond to the substrate of the **oxidase** within a potential range essentially free of interfering electrochem. reactions. Examples of glucose, alc. and amino acid sensors are shown.

- ST biosensor amperometric coimmobilized **peroxidase oxidase**; glucose sensor coimmobilized **peroxidase oxidase**; alc sensor coimmobilized **peroxidase oxidase**; amino acid sensor coimmobilized **peroxidase oxidase**
- IT Biosensors  
(amperometric, based on coimmobilized **peroxidase** and **oxidase** for alcs. and amino acids and glucose)
- IT Alcohols, analysis  
Amino acids, analysis  
RL: ANT (Analyte); ANST (Analytical study)  
(detn. of, using amperometric sensor based on coimmobilized **peroxidase** and **oxidase**)
- IT Carbon fibers, uses  
RL: ANST (Analytical study); USES (Uses)  
(hydrogen **peroxidase** immobilized on, in hydrogen peroxide amperometric sensor for anal.)
- IT Electrodes  
(amperometric, paste, **peroxidase** and **oxidase** coimmobilized on, for alcs. and amino acids and glucose detn.)
- IT Carbon fibers, uses  
RL: ANST (Analytical study); USES (Uses)  
(graphite, hydrogen **peroxidase** immobilized on Polycarbon LGR, in hydrogen peroxide amperometric sensor for anal.)
- IT 9073-63-6, Alcohol **oxidase**  
RL: ANST (Analytical study)  
(alc. amperometric biosensor based on coimmobilized horseradish **peroxidase** and, for detn. of alcs.)
- IT 9000-89-9, L-Amino acid **oxidase**  
RL: ANST (Analytical study)  
(amino acid amperometric biosensor based on coimmobilized **peroxidase** and, for anal.)
- IT 7440-44-0 7782-42-5  
RL: ANST (Analytical study)  
(carbon fibers, graphite, hydrogen **peroxidase** immobilized on Polycarbon LGR, in hydrogen peroxide amperometric sensor for anal.)
- IT 7440-44-0  
RL: ANST (Analytical study)  
(carbon fibers, hydrogen **peroxidase** immobilized on, in hydrogen peroxide amperometric sensor for anal.)
- IT 63-91-2, L-Phenylalanine, analysis 64-17-5, Ethanol, analysis 67-56-1, Methanol, analysis  
RL: ANT (Analyte); ANST (Analytical study)  
(detn. of, amperometric biosensor based on coimmobilized **peroxidase** and **oxidase** for)
- IT 7722-84-1, Hydrogen peroxide, analysis  
RL: ANT (Analyte); ANST (Analytical study)  
(detn. of, by amperometric biosensor based on immobilized **peroxidase**)
- IT 50-99-7, Glucose, analysis  
RL: ANT (Analyte); ANST (Analytical study)  
(detn. of, by using amperometric biosensor based on coimmobilized **peroxidase** and **oxidase**)

IT 9001-37-0, Glucose oxidase  
RL: ANST (Analytical study)  
(glucose amperometric sensor contg. coimmobilized peroxidase  
and, for anal.)

IT 9003-99-0, Peroxidase  
RL: ANST (Analytical study)  
(horseradish, amperometric biosensor based on  
coimmobilized oxidase and, for detn. of alcs. and amino acids  
and glucose)

IT 9002-98-6  
RL: ANST (Analytical study)  
(in amperometric biosensor based on coimmobilized  
peroxidase and oxidase, for anal.)

IT 25667-98-5, Poly-o-phenylenediamine  
RL: ANST (Analytical study)  
(in amperometric biosensor based on immobilized  
peroxidase and oxidase)

IT 51774-88-0  
RL: ANST (Analytical study)  
(in amperometric biosensor based on immobilized  
peroxidases and oxidase)

IT 111-30-8, Glutaraldehyde 151-51-9, Carbodiimide  
RL: ANST (Analytical study)  
(in immobilization of peroxidase and oxidase in  
carbon paste electrode in prepn. of amperometric  
sensors)

IT 126851-11-4, AQ 29D  
RL: ANST (Analytical study)  
(membrane, in hydrogen peroxide amperometric biosensor based  
on immobilized peroxidase)

IT 7440-44-0 7782-42-5  
RL: ANST (Analytical study)  
(carbon fibers, graphite, hydrogen  
peroxidase immobilized on Polycarbon LGR, in hydrogen peroxide  
amperometric sensor for anal.)

RN 7440-44-0 HCPLUS  
CN Carbon (7CI, 8CI, 9CI) (CA INDEX NAME)

C

RN 7782-42-5 HCPLUS  
CN Graphite (8CI, 9CI) (CA INDEX NAME)

C

IT 7440-44-0  
RL: ANST (Analytical study)  
(carbon fibers, hydrogen peroxidase  
immobilized on, in hydrogen peroxide amperometric sensor for anal.)

RN 7440-44-0 HCPLUS  
CN Carbon (7CI, 8CI, 9CI) (CA INDEX NAME)

C

IT 9003-99-0, Peroxidase  
RL: ANST (Analytical study)  
(horseradish, amperometric biosensor based on

coimmobilized oxidase and, for detn. of alcs. and amino acids  
and glucose)

RN 9003-99-0 HCAPLUS

CN Peroxidase (9CI) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

L77 ANSWER 33 OF 33 HCAPLUS COPYRIGHT 2003 ACS

AN 1991:534332 HCAPLUS

DN 115:134332

TI Amino oxidase amperometric biosensor for  
polyamines

AU Gasparini, Roberta; Scarpa, Marina; Di Paolo, Maria Luisa; Stevanato,  
Roberto; Rigo, Adelio

CS Dep. Biol. Chem., Padua Univ., Padua, 35100, Italy

SO Bioelectrochemistry and Bioenergetics (1991), 25(2), 307-15  
CODEN: BEBEBP; ISSN: 0302-4598

DT Journal

LA English

CC 17-1 (Food and Feed Chemistry)

AB An improved amino oxidase enzyme electrode  
was constructed and applied to the detn. of the amt. of polyamines present  
in real samples. The electrode is based on the amperometric  
detection of H<sub>2</sub>O<sub>2</sub> produced in the enzymic oxidn. of polyamines by  
amino oxidase. Amino-oxidase from  
soybean seedlings, characterized by an extremely high activity for  
cadaverine and putrescine, was used. The enzyme was immobilized in an  
agarose matrix in the presence of glutaraldehyde and bovine serum albumin  
on the surface of a Pt electrode. Cadaverine, in  
concns. between 0.5 and 500 .mu.M, can be quant. detd. by use of the  
amino oxidase electrode, the linear  
calibration range being 0.5-10 .mu.M. The lower detection limit was 0.2  
.mu.M and the response time was 15-60 s. Putrescine showed similar  
behavior. The max. current response for cadaverine was 5.1 .mu.A/cm<sup>2</sup>,  
with an apparent Km' of 0.175 mM. The sensor response was  
stable for >32 h of continuous operation at room temp. and, in the  
presence of fish or meat homogenates, no change in the signal-to-noise  
ratio was obsd. The long-term stability, pH, and temp. response of the  
biosensor also were studied.

ST polyamine detn enzyme bioelectrode; amine  
oxidase electrode polyamine detn

IT Michaelis constant

(of amino oxidase immobilized on amperometric  
bioelectrode)

IT Immobilization, biochemical  
(of amino oxidase, in agarose matrix on  
platinum electrode surface, polyamine detn. in  
relation to)

IT Fish

(polyamines detn. in tissue homogenates of, with amino  
oxidase amperometric bioelectrode)

IT Food analysis

(polyamines detn. in, of tissue homogenates with amino  
oxidase amperometric bioelectrode)

IT Electrodes

(bio-, enzyme, amperometric, hydrogen peroxide-selective, with  
immobilized amino oxidase, for polyamine detn.,  
characterization of)

IT Amines, analysis

RL: ANT (Analyte); ANST (Analytical study)

(poly-, detn. of, in tissue homogenates with amino  
oxidase amperometric bioelectrode)

IT Meat

(veal, polyamines detn. in tissue homogenates of, with **amino oxidase** amperometric **bioelectrode**)  
IT 7722-84-1, Hydrogen peroxide, analysis  
RL: ANT (Analyte); ANST (Analytical study)  
(detn. of, by **amino oxidase** enzyme  
**electrode**, polyamine detn. in relation to)  
IT 71-44-3, Spermine 110-60-1, Putrescine 124-20-9, Spermidine  
462-94-2, Cadaverine  
RL: ANT (Analyte); ANST (Analytical study)  
(detn. of, in tissue homogenates with **amino oxidase**  
amperometric **electrode**)  
IT 9000-89-9  
RL: BIOL (Biological study)  
(immobilized, in agarose matrix on **platinum electrode**  
, for polyamine detn. in tissue homogenates)

=> d his

(FILE 'HOME' ENTERED AT 10:01:49 ON 17 JAN 2003)  
SET COST OFF

FILE 'HCAPLUS' ENTERED AT 10:02:04 ON 17 JAN 2003

	E CSOREGI E/AU
L1	48 S E3-E5
	E MICULESCU M/AU
	E NICULESCU M/AU
L2	26 S E3,E12
	E FREBORT I/AU
L3	68 S E3,E4
	E WO2000-SE1449/AP, PRN
L4	1 S E3,E4
	E SE99-2608/AP, PRN
L5	1 S E4
L6	1 S L1-L3 AND L4,L5
	E FORSKARPATENT/PA, CS
L7	26 S E3-E20
L8	16015 S BIOSENSOR OR BIO SENSOR
L9	32 S L1-L3 AND L8
L10	3 S L7 AND L8

FILE 'REGISTRY' ENTERED AT 10:05:26 ON 17 JAN 2003

L11	1 S 9003-99-0
L12	1 S 9059-11-4
	E AMINE OXIDASE/CN
	E PEROXIDASE/CN
L13	1 S L11 AND ?PEROXIDASE?/CNS
L14	1 S L12 AND (AMINE(L)OXIDASE)/CNS

FILE 'HCAPLUS' ENTERED AT 10:07:22 ON 17 JAN 2003

L15	1848 S L14
L16	3429 S AMINEOXIDASE OR AMINE OXIDASE
L17	30631 S L13
L18	77352 S ?PEROXIDASE?
L19	3448 S L15,L16
L20	78099 S L17,L18
	E BIOSENSOR/CT
L21	4526 S E5-E29
	E E4+ALL
L22	10181 S E7
	E E6+ALL
L23	21517 S E4,E5
L24	17 S L19 AND L8

L25 11 S L19 AND L21-L23  
L26 6 S L24, L25 AND L20  
L27 19 S L24, L25, L26  
L28 6 S L6, L9, L10 AND L27  
L29 28 S L6, L9, L10 NOT L28  
L30 1 S L27 AND REDOX(L) HYDROGEL  
L31 6 S L29 AND REDOX(L) HYDROGEL  
L32 9 S L27 AND (GOLD OR SILVER OR PLATINUM OR PALLADIUM OR COPPER OR  
L33 19 S L29 AND (GOLD OR SILVER OR PLATINUM OR PALLADIUM OR COPPER OR

FILE 'REGISTRY' ENTERED AT 10:12:20 ON 17 JAN 2003  
L34 6 S (PALLADIUM OR PLATINUM OR SILVER OR GOLD OR CARBON OR GRAPHIT

FILE 'HCAPLUS' ENTERED AT 10:12:37 ON 17 JAN 2003  
L35 5 S L34 AND L27  
L36 9 S L34 AND L29  
L37 4 S L27 AND (OSMIUM OR OS)  
L38 6 S L29 AND (OSMIUM OR OS)

FILE 'REGISTRY' ENTERED AT 10:14:00 ON 17 JAN 2003  
L39 1 S 25232-42-2

FILE 'HCAPLUS' ENTERED AT 10:14:54 ON 17 JAN 2003  
L40 2 S L39 AND L27  
L41 3 S L39 AND L29

FILE 'REGISTRY' ENTERED AT 10:15:46 ON 17 JAN 2003  
L42 1 S 26403-72-5  
L43 1 S 7440-04-2

FILE 'HCAPLUS' ENTERED AT 10:15:57 ON 17 JAN 2003  
L44 2 S (L42 OR L43) AND L27  
L45 4 S (L42 OR L43) AND L29  
L46 19 S L27, L28, L30, L32, L35, L40, L44

FILE 'REGISTRY' ENTERED AT 10:16:53 ON 17 JAN 2003  
L47 1 S 7440-50-8

FILE 'HCAPLUS' ENTERED AT 10:16:59 ON 17 JAN 2003  
L48 2 S L47 AND L27  
L49 5 S L47 AND L29  
L50 19 S L46, L48

FILE 'REGISTRY' ENTERED AT 10:18:26 ON 17 JAN 2003  
L51 1 S 1134-35-6

FILE 'HCAPLUS' ENTERED AT 10:18:33 ON 17 JAN 2003  
L52 1 S L51 AND L27  
L53 1 S L51 AND L29  
L54 19 S L50, L52  
L55 19 S L54 AND (?SENSOR? OR ?ELECTRODE? OR ?OXIDASE? OR AMIN# OXIDAS  
L56 4 S L55 AND ?GRAPHITE?  
L57 5 S L55 AND (C OR CARBON)  
L58 8 S L56, L57  
L59 8 S L55 AND ?POLYM?  
L60 11 S L58, L59  
L61 3 S L55 AND PEA  
L62 19 S L55, L61  
SEL DN AN 1 2  
L63 17 S L62 NOT E1-E6  
L64 23 S L31, L33, L36, L38, L41, L45, L49, L53  
L65 5 S L29 NOT L62, L64  
SEL DN AN 1 3

L66 3 S L65 NOT E7-E12  
L67 20 S L63, L66  
SEL DN AN 1 3 6 8 13 14 15 16 17  
DEL SEL  
SEL DN AN 1 3 6 8 13 14 15 16 17 L64  
L68 14 S L64 NOT E1-E25  
SEL DN AN 10  
L69 13 S L68 NOT E26-E28  
L70 33 S L67, L69 AND L1-L10, L15-L33, L35-L38, L40, L41, L44-L46, L48-L50, L5  
L71 29 S L70 AND (PEA OR SWEET(L) POTATO OR HORSERADISH OR SOYBEAN OR S  
L72 15 S L70 AND (?POLYM? OR POLY ETHYLENEGLYCOL DIGLYCIDYL ETHER OR V  
L73 4 S L70 AND (POLY ETHYLENE GLYCOL DIGLYCIDYL ETHER)  
L74 33 S L70-L73  
L75 7 S L74 AND (C OR CARBON) (L) (PASTE# OR FIBER OR FIBRE OR VITROUS)  
L76 9 S L74 AND GRAPHITE  
L77 33 S L74-L76

FILE 'HCAPLUS' ENTERED AT 10:35:52 ON 17 JAN 2003